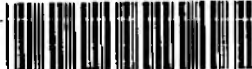


GRAPHICAL REPRESENTATIONS OF CERTAIN METALS AND  
ORGANICS CONCENTRATIONS IN RELATION TO FLOW  
RATES AND SAMPLING TIMES FOR SOME OF THE RIVERS  
IN DISTRICT 6

110

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ENVIRONMENT AGENCY



042456

## INTRODUCTION

The aim of this report is to establish a link between certain metals and organic compounds and the factors that may influence their concentration. The factors considered are flow rate and the time of sampling for the river Wandle, the Beverley Brook, the Ravensbourne and the tidal Thames.

The data used has been collated over the period January 1986 to September 1990.

## RESULTS DESCRIPTION

### 1. Variation of concentrations with flow in the Beverley Brook at Priests Bridge

#### 1.1 Zinc concentrations variation with flow (Fig.1)

From the results the majority of samples were taken at flows between  $0.3 \text{ m}^3 \text{ s}^{-1}$  to  $1.1 \text{ m}^3 \text{ s}^{-1}$  and Zinc concentrations of  $0.033 \text{ ug l}^{-1}$  to  $0.103 \text{ ug l}^{-1}$ . Of the three samples taken at times of flow greater than  $1.2 \text{ m}^3 \text{ s}^{-1}$  there appears to be no relationship between zinc concentration and flow, reflecting the pattern of the results overall.

#### 1.2 HCH $\gamma$ concentration variation with flow (Fig.2)

Sampling took place at times of relatively low flow ( $0.03$  to  $1.1 \text{ m}^3 \text{ s}^{-1}$ ) where HCH concentrations varied between  $0.035$  to  $0.2 \text{ ug l}^{-1}$ . However, there appears to be no obvious affiliation between the parameters.

#### 1.3 Dieldrin concentration variation with flow (Fig.3)

Over the period 9.2.88 to 8.8.90, 21 samples were analysed for dieldrin concentration. The flow rates at the times of sampling fluctuated between  $0.4$  to  $0.13 \text{ m}^3 \text{ s}^{-1}$ . Dieldrin concentrations loosely increased with the flow rate from  $0.001$  to  $0.02 \text{ ug l}^{-1}$ .

#### 1.4 Trichloroethene concentration variation with flow (Fig.4)

All of the nine samples taken for Trichloroethene analysis from 11.12.89 to 8.8.90 were taken when the flow rate ranged from  $0.3$  to  $0.86 \text{ m}^3 \text{ s}^{-1}$ . The trichloroethene concentrations varied from  $0.1$  to  $0.3 \text{ ug l}^{-1}$  with one exceptionally high value at  $4.2 \text{ ug l}^{-1}$ . Overall there appears to be no obvious relationship between flow and concentration.

## 2. Variation of concentration with flow in the Ravensbourne at Deptford Bridge

### 2.1 Zinc concentration variation with flow (Fig.5)

The majority of the 23 samples taken for zinc analysis were taken at times of relatively low flow ( $0.1$  to  $0.55 \text{ m}^3 \text{ s}^{-1}$ ) where zinc concentration ranged from  $0.01$  to  $0.054 \text{ mg l}^{-1}$  in no particular arrangement. The two samples taken at flows of  $0.61$  and  $1.14 \text{ m}^3 \text{ s}^{-1}$  had zinc concentrations of  $0.05$  and  $0.06 \text{ mg l}^{-1}$  respectively. Overall the results appear to be of a random nature.

### 2.2 HCH Y concentration variation with flow (Fig.6)

From the 19 HCH Y samples taken, most occur in the  $0.1$  to  $0.61 \text{ m}^3 \text{ s}^{-1}$  flow range, with only two samples taken at flows greater than this. As in the case with zinc, there appears to be little or no correlation of HCH concentration to flow rate. However, the maximum concentration value of  $0.077 \text{ ug l}^{-1}$  occurs at  $0.14 \text{ m}^3 \text{ s}^{-1}$  and the joint minimum value of  $0.001 \text{ ug l}^{-1}$  at  $1.14 \text{ m}^3 \text{ s}^{-1}$ . Most of the concentrations of HCH Y at the low flows are less than  $0.03 \text{ ug l}^{-1}$ .

### 2.3 Dieldrin concentration variation with flow (Fig.7)

Dieldrin concentrations vary between  $0.001$  and  $0.005 \text{ ug l}^{-1}$  and appear unaffected by flow rates, remaining random in their distribution.

### 2.4 Trichloroethene concentration variation with flow (Fig.8)

Out of the eight samples taken, seven have concentrations of  $0.1 \text{ ug l}^{-1}$  at flow rates varying between  $0.15$  to  $0.62 \text{ m}^3 \text{ s}^{-1}$ . The only "outlier" result is a sample with a concentration of  $0.3 \text{ ug l}^{-1}$  at a flow rate of  $0.13 \text{ m}^3 \text{ s}^{-1}$ .



### 3. Variation of concentration with flow in the Thames at Teddington

#### 3.1 Zinc concentration variation with flow (Fig.9)

The trend in zinc concentration with flow is one of an even distribution of concentration values throughout the range of flows. The majority of samples had concentrations varying between 0.01 to 0.03  $\text{mg l}^{-1}$ , but at both ends of the flow range (4 - 334  $\text{m}^3 \text{s}^{-1}$ ) higher values did occur.

#### 3.2 HCH $\gamma$ concentration variation with flows (Fig.10)

The flow rates when sampling fluctuated between 4 to 334  $\text{m}^3 \text{s}^{-1}$  and the majority of samples were taken when flows were less than 100  $\text{m}^3 \text{s}^{-1}$ . HCH  $\gamma$  concentrations of the samples taken when flows were less than 100  $\text{m}^3 \text{s}^{-1}$ , varied mainly between 0.01 and 0.055  $\text{ug l}^{-1}$ , however there were five samples which had concentrations exceeding 0.055  $\text{ug l}^{-1}$ , the largest of which was 0.306  $\text{ug l}^{-1}$ . Samples taken when flows exceeded 100  $\text{m}^3 \text{s}^{-1}$  showed far less variation in concentration (0.005 to 0.024  $\text{ug l}^{-1}$ ) and less samples were taken under these conditions.

#### 3.3 Dieldrin concentration variation with flow (Fig.11)

Dieldrin concentrations remained constant with changes in flow, at a level of 0.005  $\text{ug l}^{-1}$ . However, lower values, to a minimum of 0.001  $\text{ug l}^{-1}$  were also recorded and their distribution was mainly random in relation to the different flow rates.

#### 3.4 Trichloroethene concentration variation flow (Fig.12)

Trichloroethene concentration varied between 0.7 to 14  $\text{ug l}^{-1}$  but values are mostly in the 1 - 3  $\text{ug l}^{-1}$  range. The highest values of concentration occur at the lowest flows when most of the samples were taken. The three samples taken, when the flow rate exceeded 40  $\text{m}^3 \text{s}^{-1}$ , had values between 1 and 2  $\text{ug l}^{-1}$ .

#### 4. Variation of concentration with flow in the Wandle at the Causeway

##### 4.1 Zinc concentration variation with flow (Fig.13)

The majority of sampling occurred during times of relatively low flow (between 1 and 3 cumecs), where zinc concentrations ranged between 0.003 to 0.143  $\text{mg l}^{-1}$  plus two exceptionally high values of 0.493 and 0.465  $\text{mg l}^{-1}$ . Only 7 of the 59 samples were taken when flows exceeded 3 cumecs with the zinc values ranging from 0.01 to 0.053  $\text{mg l}^{-1}$ , plus one exceptionally high result of 0.29  $\text{mg l}^{-1}$ .

##### 4.2 HCH Y concentration variation with flow (Fig.14)

From the 29 samples taken between 14.12.87 and 8.8.90, 25 were taken between flow rates of 1.1 and 2.3  $\text{m}^3 \text{s}^{-1}$  and only 4 samples at flows greater than 2.3  $\text{m}^3 \text{s}^{-1}$ . The range of HCH Y concentrations obtained at flows between 1.1 and 2.3  $\text{m}^3 \text{s}^{-1}$  varied between 0.001 to 0.2  $\text{ug l}^{-1}$ , plus one exceptional value of 0.683  $\text{ug l}^{-1}$  at 1.12  $\text{m}^3 \text{s}^{-1}$ . HCH Y concentration values at flows greater than 2.3  $\text{m}^3 \text{s}^{-1}$  deviated between 0.085  $\text{ug l}^{-1}$  (at 3.64  $\text{m}^3 \text{s}^{-1}$ ) and 0.15  $\text{ug l}^{-1}$  (at 2.76  $\text{m}^3 \text{s}^{-1}$ ), concentration therefore decreases with an increase in flow.

##### 4.3 Dieldrin concentration variation with flow (Fig.15)

During the period 21.3.88 to 8.8.90, 21 samples were taken for Dieldrin analysis. The majority of sampling (19 samples) occurred at times of relatively low flow (1.2 - 2.24  $\text{m}^3 \text{s}^{-1}$ ) and only two samples taken at greater flows (3.28 - 3.64  $\text{m}^3 \text{s}^{-1}$ ). The Dieldrin concentration values at the flows between 1.2 and 2.24  $\text{m}^3 \text{s}^{-1}$  fluctuated between 0.001 to 0.018  $\text{ug l}^{-1}$  with no apparent relationship between flow and Dieldrin concentration. At the greater flows the minimum result was 0.003  $\text{ug l}^{-1}$  and at the lower flows the maximum result was 0.005  $\text{ug l}^{-1}$ .

##### 4.4 Trichloroethene concentration variation with flow (Fig.16)

From the results it appears that Trichloroethene concentration increases with flow, the lowest value of 0.1  $\text{ug l}^{-1}$  is recorded at flows of between 1.1 and 1.5 cumecs with one of the highest values of 0.8  $\text{ug l}^{-1}$  occurring

at 2.24 cumecs. There is one high "outlier" value of  $2.0 \text{ ug l}^{-1}$  at 1.71 cumecs.

## 5. Variation of concentration with sampling time

### 5.1 Variation of Nickel concentration with sampling time

In the Wandle at the Causeway (Fig.17)

All sampling took place between 10.00 hours and 16.30 hours. The minimum Nickel concentration recorded was  $0.005 \text{ mg l}^{-1}$  and the maximum concentration was  $0.044 \text{ mg l}^{-1}$ . Most of the samples taken had a concentration of  $0.01 \text{ mg l}^{-1}$  and all values appeared to be independent of the time of sampling.

### 5.2 Variation in Chromium concentration with sampling time

in the Beverley Brook at Priests Bridge (Fig.18)

All of the samples were taken between 10.00 hours and 18.00 hours. The majority of the concentrations recorded were at one of either two values :  $0.005$  or  $0.01 \text{ mg l}^{-1}$ , with only one exception of  $0.009 \text{ mg l}^{-1}$ . From the graph there appears to be no direct correlation between sampling time and Chromium concentration.

### 5.3 Variation in Chromium concentration with sampling time

in the Wandle at the Causeway (Fig.19)

The lowest Chromium concentration recorded was  $0.005 \text{ mg l}^{-1}$  and the highest concentration was  $0.044 \text{ mg l}^{-1}$ . However, the majority of the samples taken had either a concentration of  $0.005 \text{ mg l}^{-1}$  or  $0.01 \text{ mg l}^{-1}$  and there appeared to be no obvious relationship between Chromium concentrations and sampling time.

### 5.4 Variation in Zinc concentration with sampling time

in the Beverley Brook at Priests Bridge (Fig.20)

The Zinc concentrations varied between the maximum of  $0.247 \text{ mg l}^{-1}$  at 11.18 hours and the minimum value of  $0.04 \text{ mg l}^{-1}$  at 10.10 hours. However, the majority of samples had concentrations ranging between  $0.04 \text{ mg l}^{-1}$  and  $0.08 \text{ mg l}^{-1}$  spread randomly through the 10.00 to 16.00 hours sampling window. The higher concentrations were recorded between 11.00 and 13.00 hours.

### 5.5 Variation in Zinc concentration with sampling time

#### In the Wandle at the Causeway (Fig.21)

All samples were taken between 10.00 and 16.30 hours, where Zinc concentrations ranged between 0.004 and 0.493  $\text{mg l}^{-1}$ . Concentration values are randomly spread throughout the sampling window and mainly vary between 0.04 and 0.11  $\text{mg l}^{-1}$ .

## 6. Variation of concentrations in the tidal Thames in relation to flows at Teddington

In this section the determinand concentrations for each zone of the tidal Thames are related to one of two flow rates taken at Teddington. The first flow rate ("low flow") relates to a flow of 20 cumecs or less for at least 7 days before the sample was taken, and the second flow rate ("high flow") relates to a flow of 40 cumecs or greater, also for at least 7 days before the sample was taken.

### 6.1 Variation in HCH $\gamma$ concentration in the Thames in relation to flow rates at Teddington (Fig.22)

As Fig.22 shows, HCH  $\gamma$  concentrations decrease as samples are taken towards the estuary. The maximum recorded value of  $0.103 \text{ ug l}^{-1}$  occurs in zone 3 and was recorded at a time of "low flow" at Teddington. The minimum concentration of  $0.002 \text{ ug l}^{-1}$  was recorded in zone 29 at a time of "high flow" at Teddington. On the whole the lowest values in each zone correspond to times of "high flow" at Teddington.

### 6.2 Variation in Atrazine concentration in the Thames in relation to flow rates at Teddington (fig.23)

Most of the samples taken were taken at times of "low flow" at Teddington. Apart from some "outliers" the maximum concentration of Atrazine in the Thames remains consistent at approximately  $0.9 \text{ ug l}^{-1}$ , but eventually starts to decrease from zone 17 to  $0.06 \text{ ug l}^{-1}$  in zone 27. Those samples taken during times of "high flow" appear randomly intermixed with the concentrations of samples taken during "low flow".

### 6.3 Variation in Simazine concentration in the Thames in relation to flow rates at Teddington (Fig.24)

Simazine concentration levels rise from zone 3 to the peak at  $1.5 \text{ ug l}^{-1}$  in zone 9. Samples taken in later zones show a decline in concentration to zone 27 where the maximum concentration recorded was  $0.09 \text{ ug l}^{-1}$ . The higher concentrations in each zone were recorded at times of "low flow" at Teddington.

#### 6.4 Variation in Cadmium concentration in the Thames in relation to flow rates at Teddington (Fig.25)

The maximum Cadmium concentration of  $0.002 \text{ mg l}^{-1}$  was recorded in zone 27, but the overall trend appears to be a gradual decline in concentration from zone 3 towards the estuary. All samples from zones 1-7 were taken at times of "low flow" at Teddington, all the samples taken further down the estuary were taken at times of "high flow" at Teddington.

#### 6.5 Variation in Copper concentration in the Thames in relation to flow rates at Teddington (Fig.26)

Essentially, two peaks in maximum zonal Copper concentrations are recorded : one at zone 9 and one at zone 18. Levels rise from  $0.015 \text{ mg l}^{-1}$  in zone 2 to  $0.031 \text{ mg l}^{-1}$  in zone 9. Concentrations then fall to  $0.01 \text{ mg l}^{-1}$  in zone 14 and increase to  $0.022 \text{ mg l}^{-1}$  in zone 18, which is followed by another decline in concentration to  $0.007 \text{ mg l}^{-1}$  in zone 29. Flow rate at Teddington appears to have no obvious effect on levels of concentration.

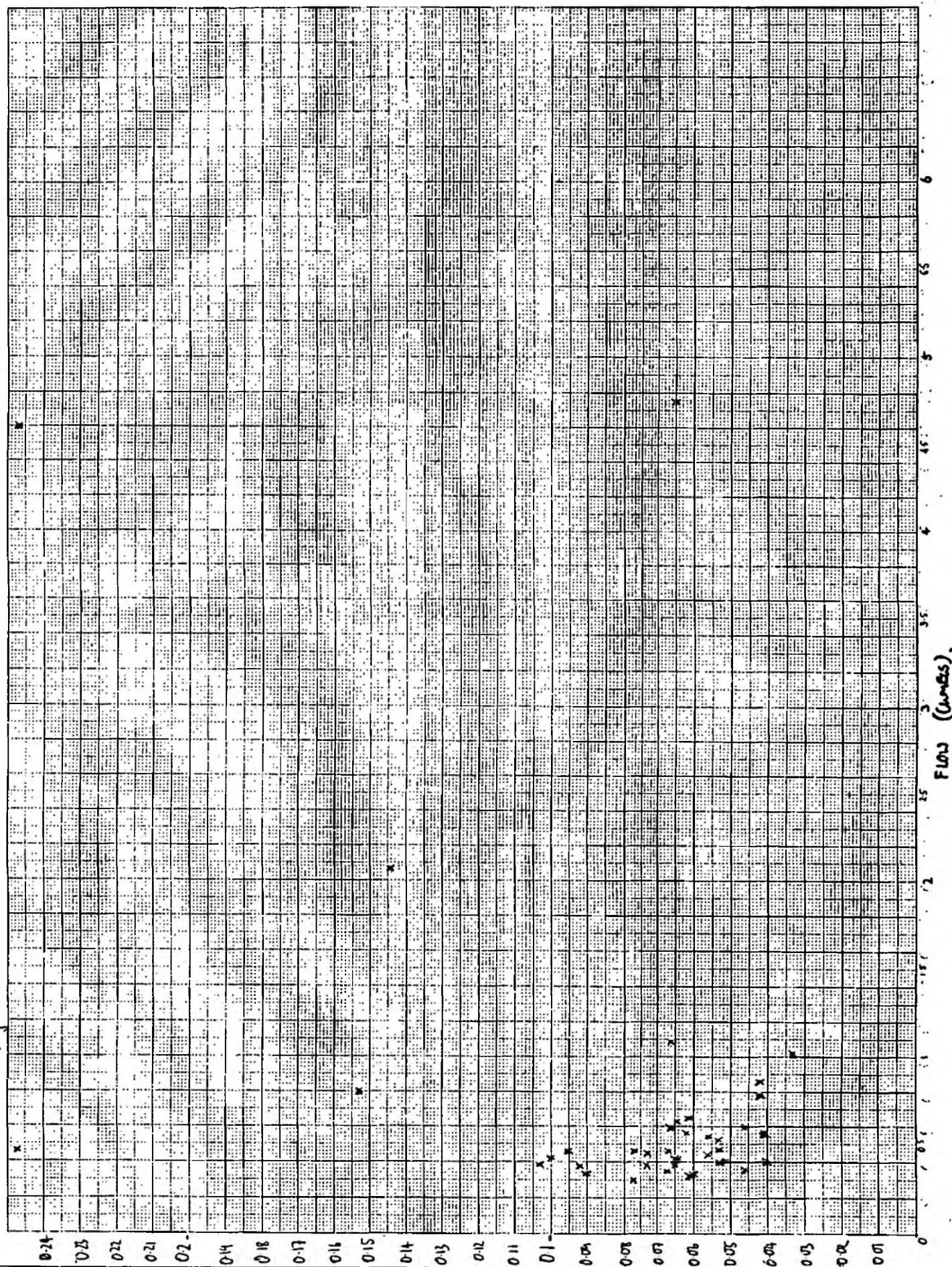
#### 6.6 Variation in Mercury concentration in the Thames in relation to flow rates at Teddington (Fig.27)

The peak in Mercury concentration is located in the central zones, levels decrease both further up and downstream. The lowest maximum concentration of  $0.15 \text{ ug l}^{-1}$  is found in zone 29. Concentration in all zones exhibit large variations between their maximum and minimum values, most minimum values are less than  $0.06 \text{ ug l}^{-1}$ . The flow rates at Teddington appear to bear no relationship with concentration.

#### 6.7 Variation in Zinc concentration in the Thames in relation to flow rates at Teddington (Fig.28)

Zinc concentration exhibits a random variation throughout all the zones with results varying between  $0.249 \text{ mg l}^{-1}$  in zone 17 and  $0.005 \text{ mg l}^{-1}$  in zones 27 and 28. However, in the earlier zones (up to 18) all the maximum concentrations were recorded at times of "low flow" at Teddington and vice versa further down the estuary.

RELATION IN ZINC CONCENTRATION WITH FLOW IN THE HOVELEY BROOK AT PUEBLO BEOME (112.88 - 12.9.90)  
(Fig 1)





VARIATION IN WCH <sup>igw</sup> WITH FLOW IN THE BOBBLEY BROOK AT MISSIS BARGE (9288-8890)  
(Fig 2)

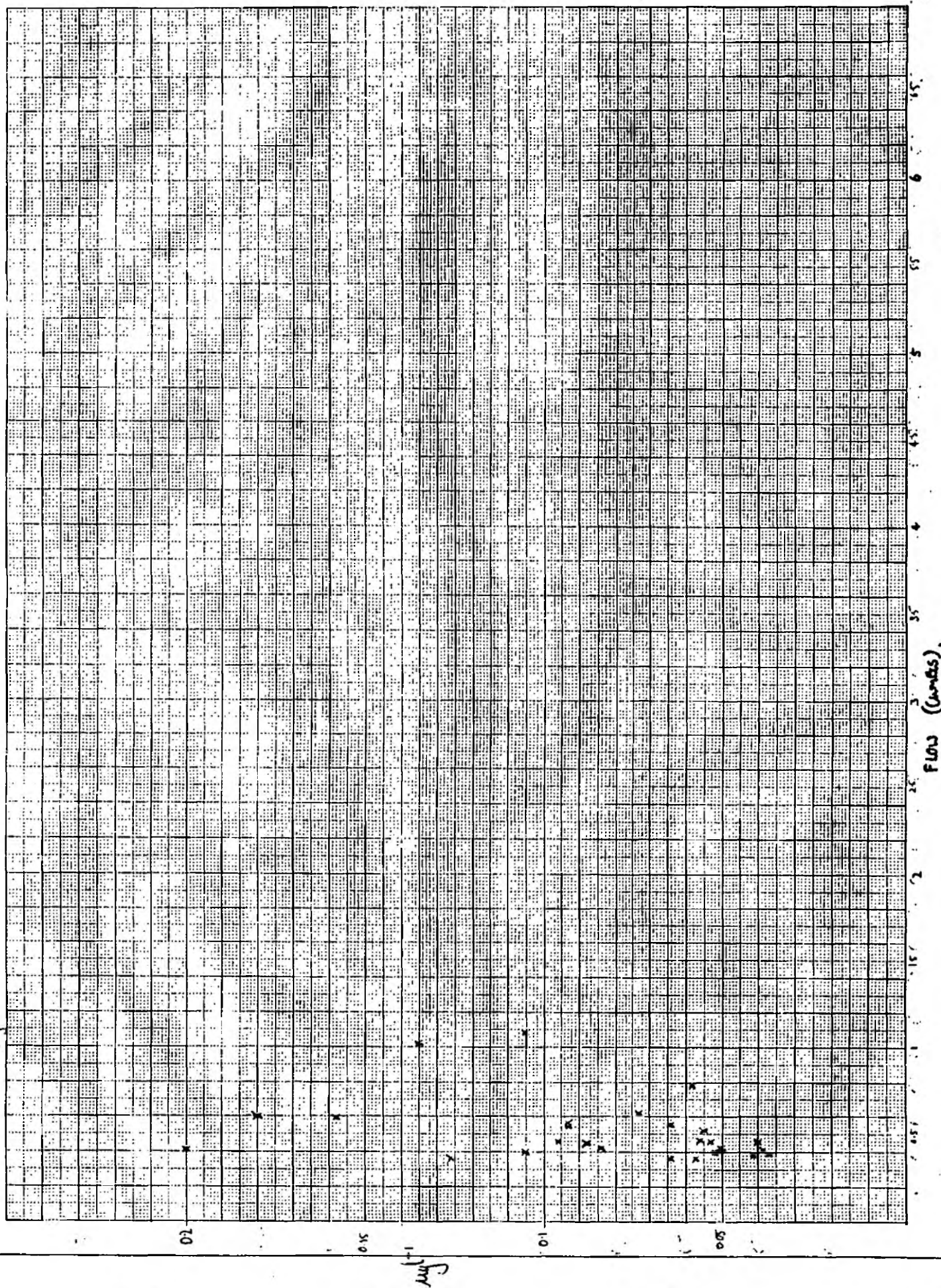
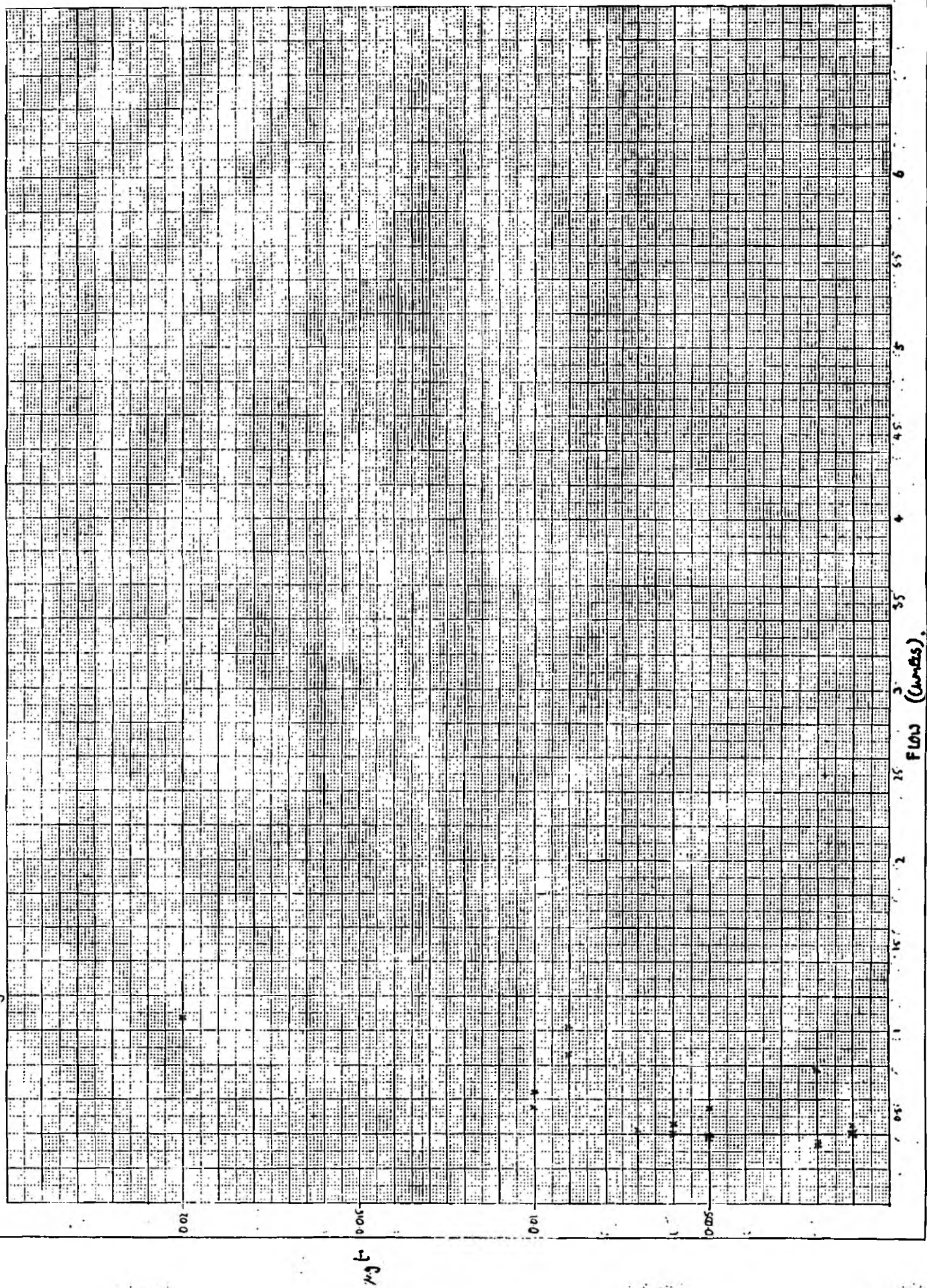
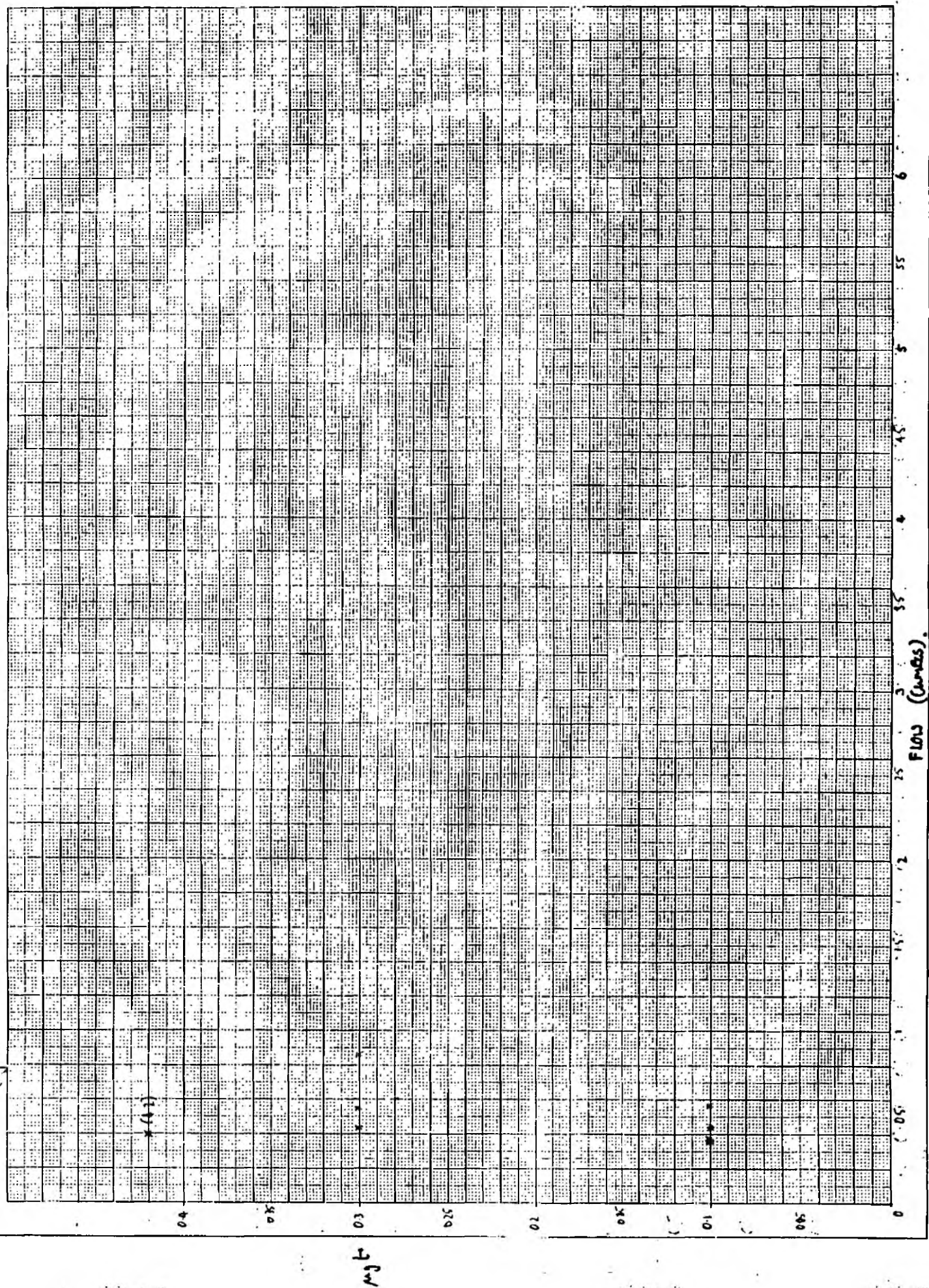


Fig. 3  
 VARIATION IN DILUTION CONCENTRATION WITH FLOW IN THE BEVERLY SEWAGE TREATMENT PLANT (4/12/58 - 8/8/60)

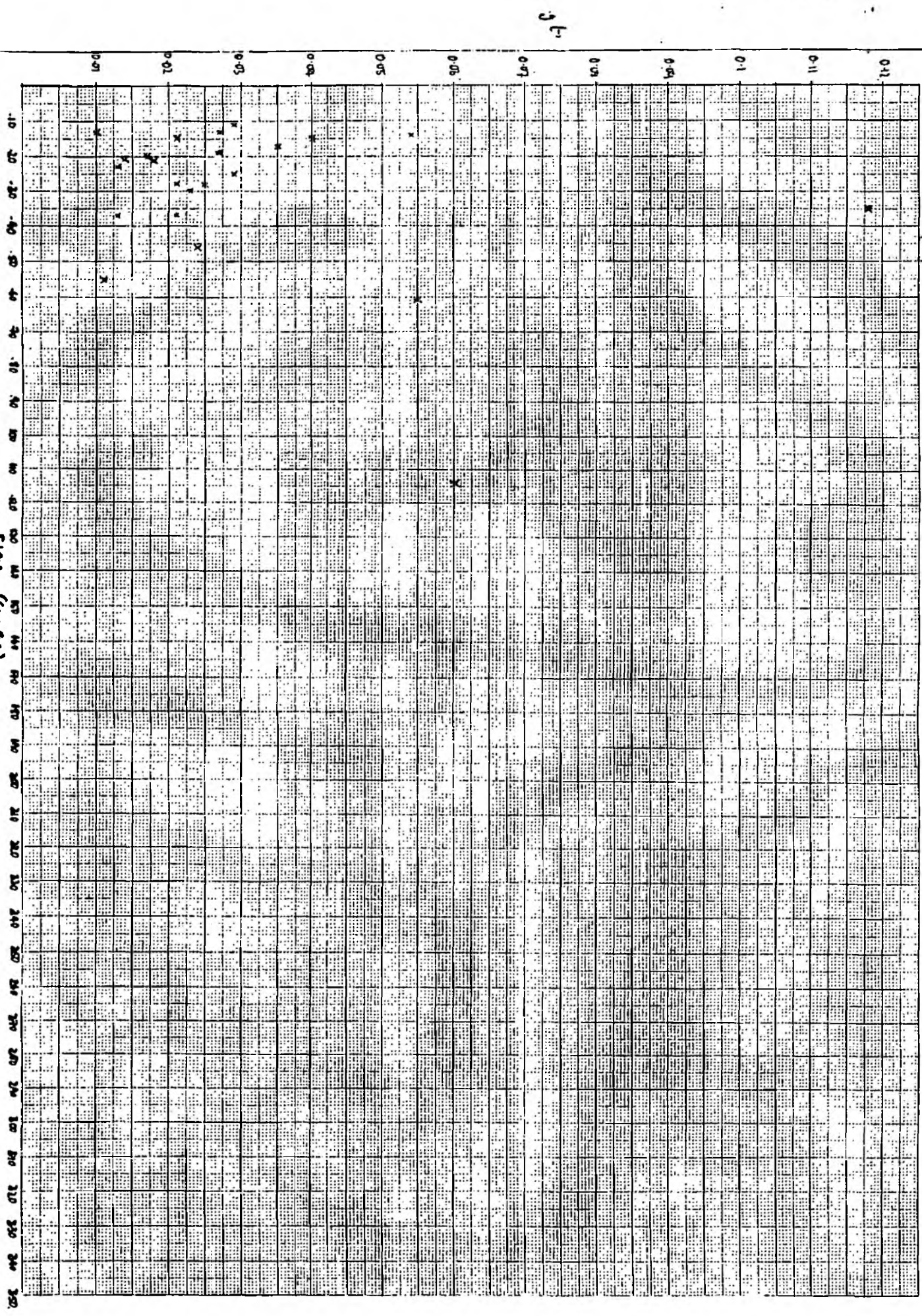




VARIAION IN TEMPERATURE (INCUBATION) WITH FLOW IN THE REVERSE ROOM AT PLOTS BRIDGE (11/12/44 - 8/8/46)  
(Fig. 4)

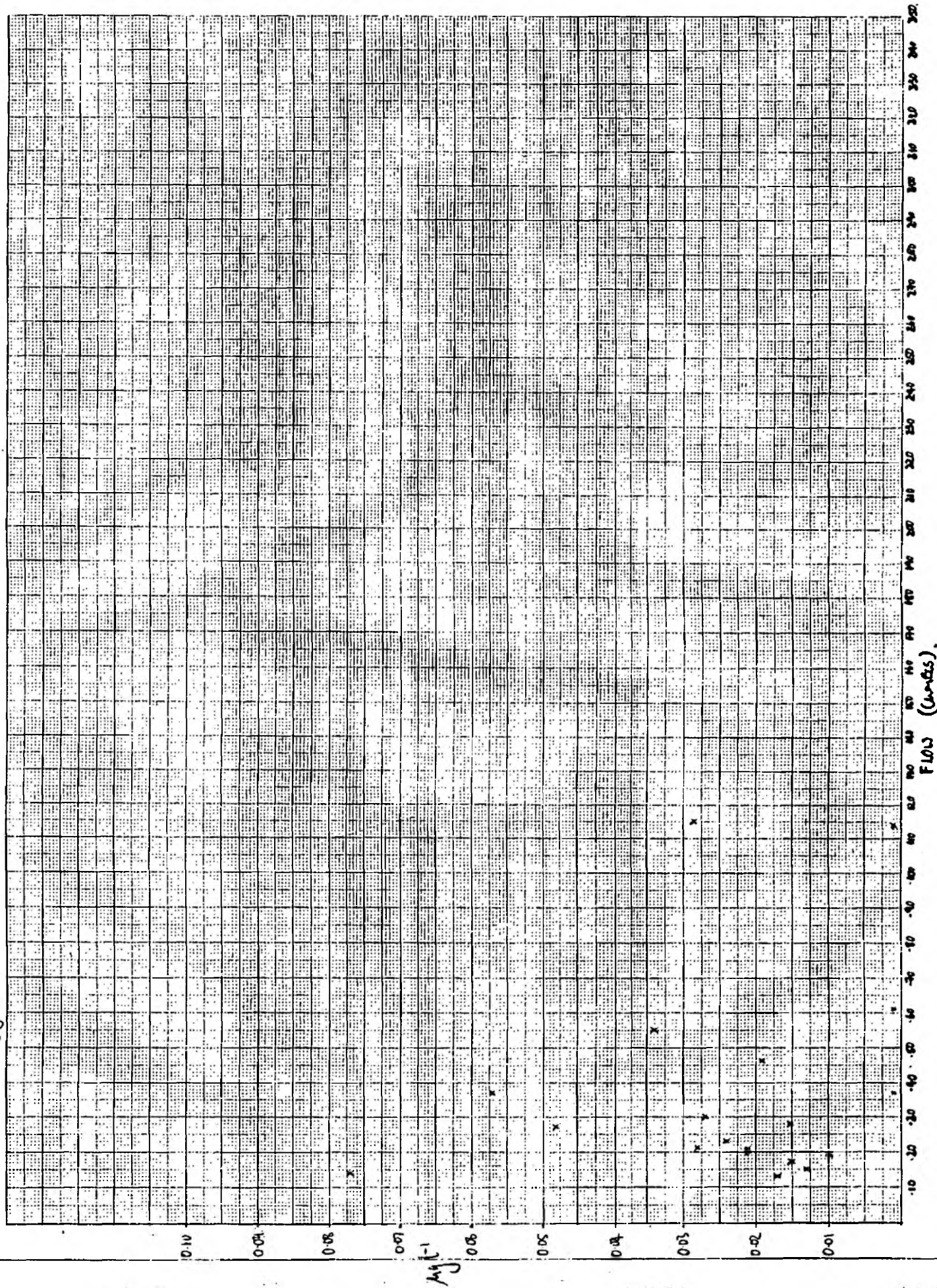


Variation of Time Lag with Flow in the Savannah at Oxford Road.  
 (Fig. 51)



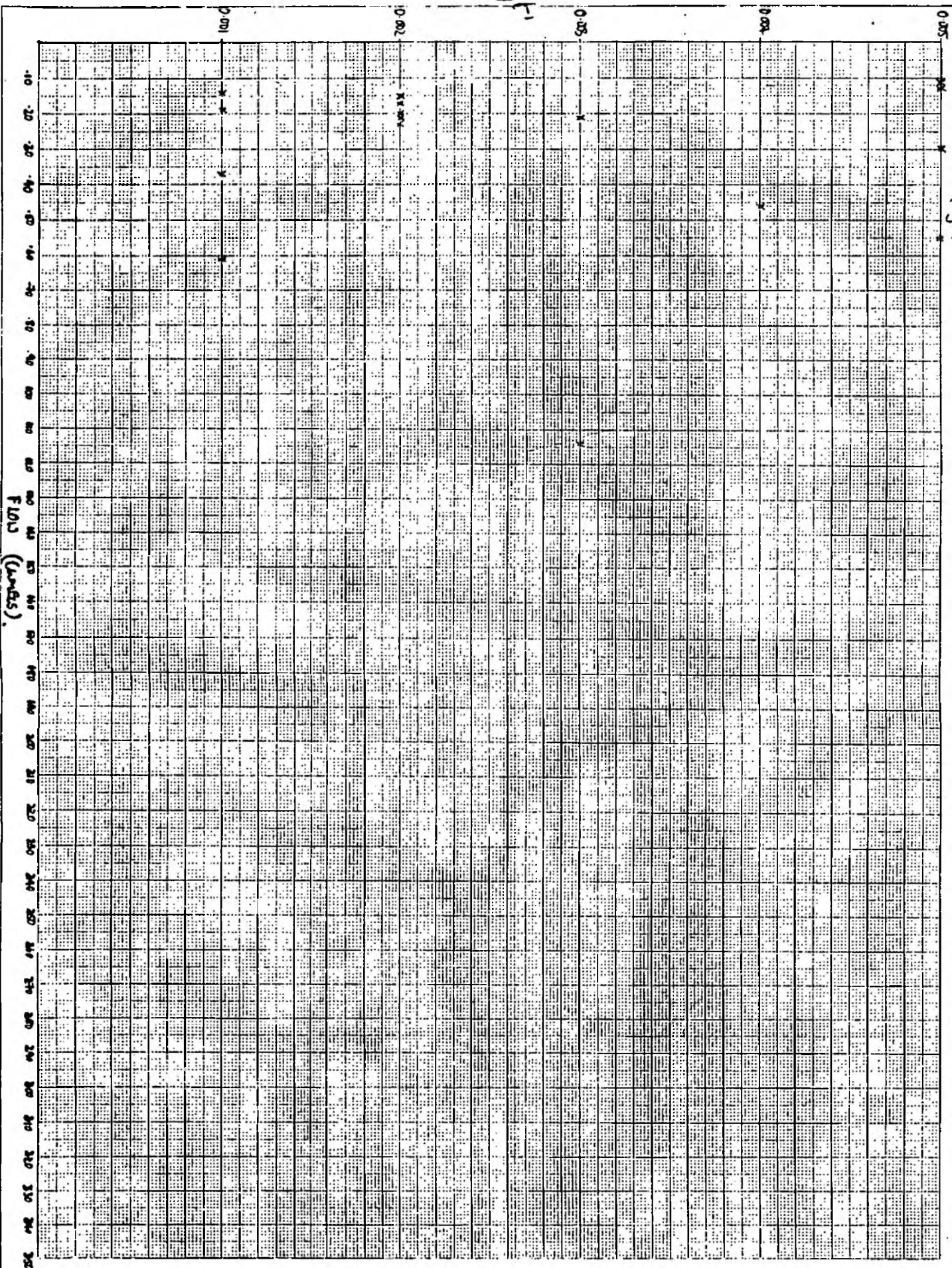


VARIATION OF  $\text{HCl}$  CONCENTRATION WITH FLOW IN THE RAYONBERG AT DORTMUND GERMANY.  
(Fig. 6)



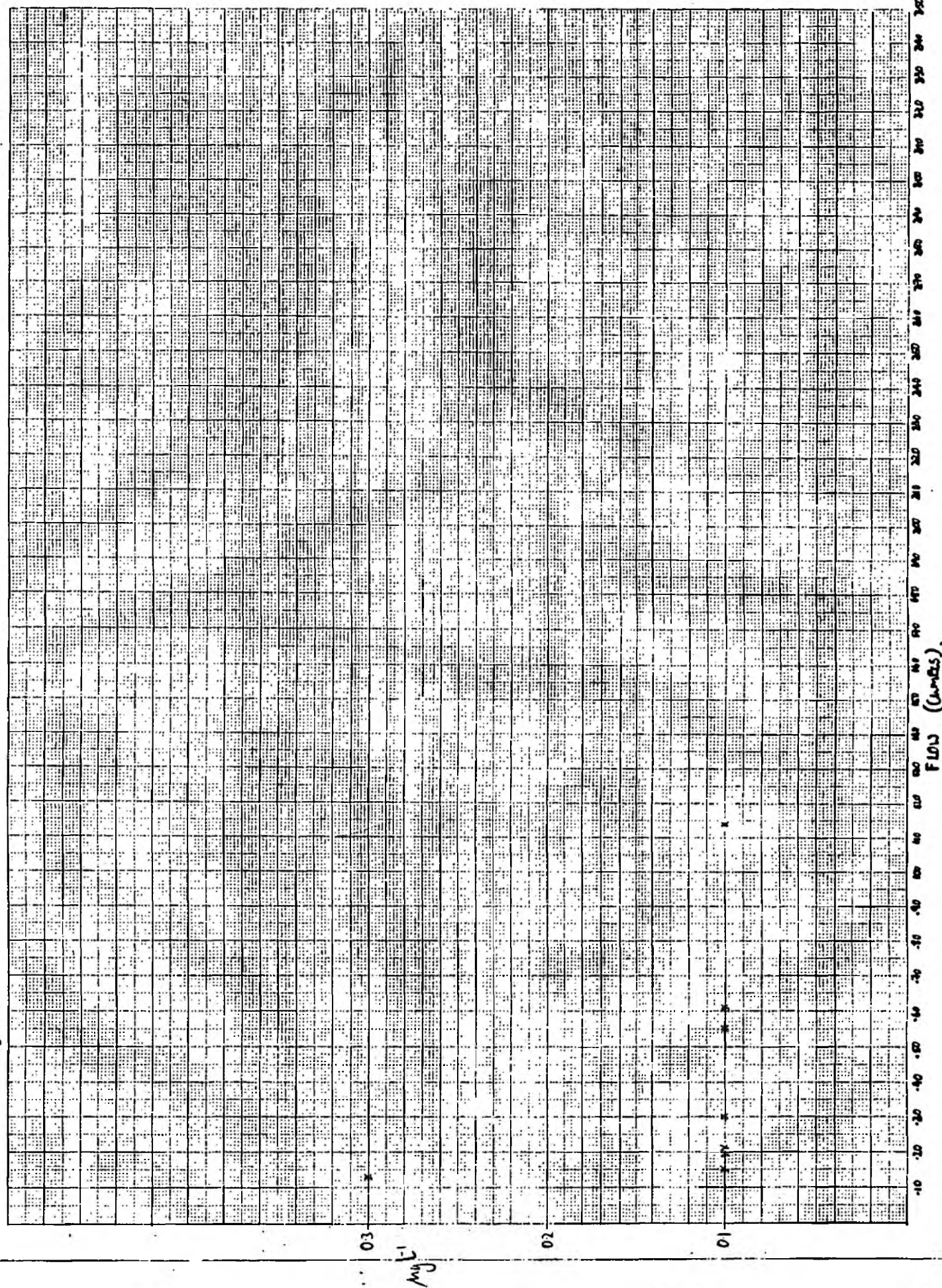
# VARIATION IN DIACON (CONCENTRATION WITH FLOW IN THE CONCENTRATION AT DIFFERENT GRADES (7-3-88 - 8-8-90)

(Fig 7)



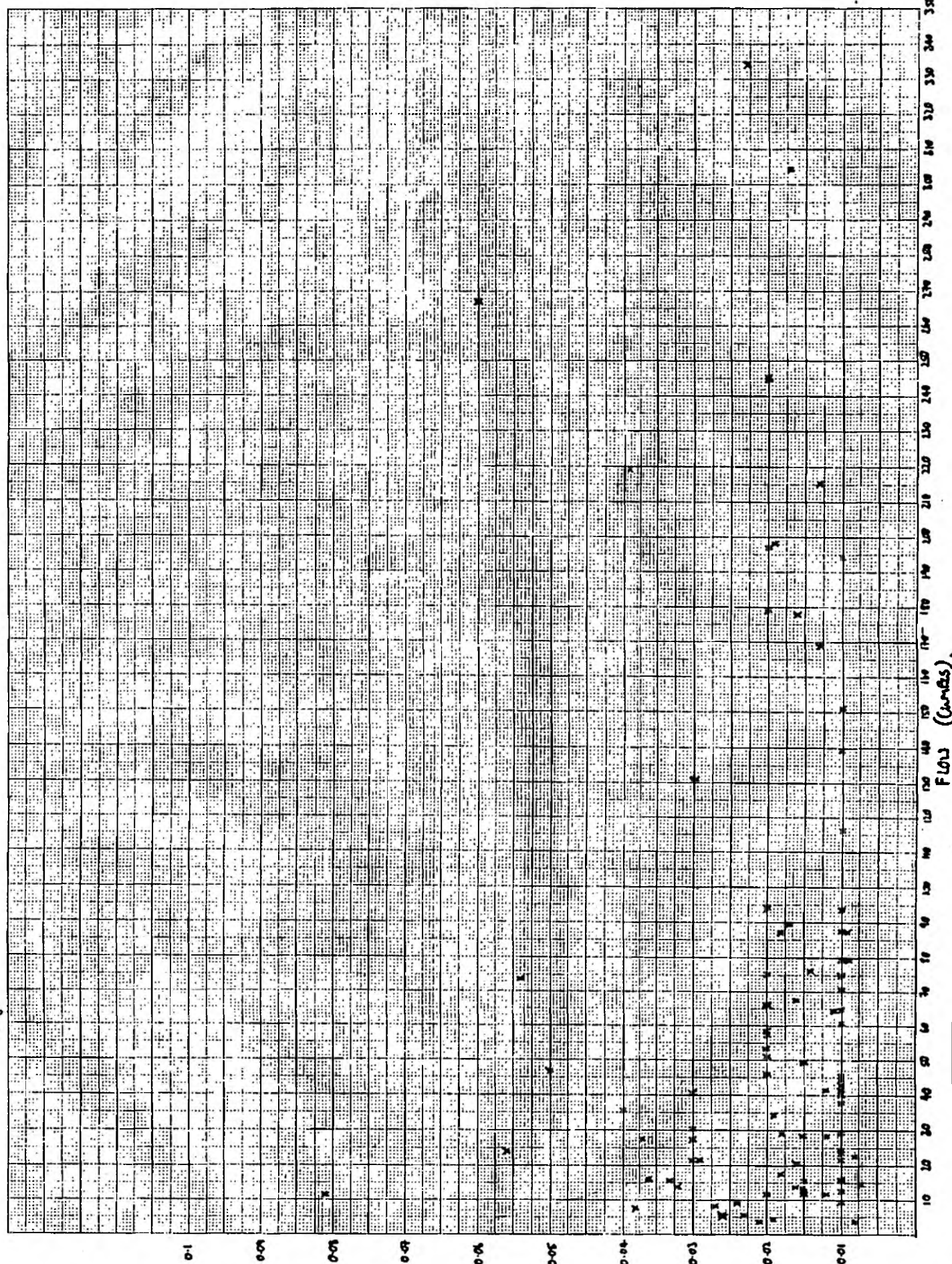


VARIATION IN TELLURIDE CONCENTRATION WITH FLOW IN THE RAVENSBURG AT DORTCH BEINE (5-12-84 - 8-8-90)  
(Fig 8)



VACCINATION OF ZINK CALCINATION WITH FLUX IN THE TURNES AT TOSBINGTON WERE (9/11/66 - 9/8/90).

(Fig. 9)



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Variation of HCH (ppm) Concentration With Flow in the Trunks at Termination LWR (9/1/80 - 9/8/90)

(Fig 10)

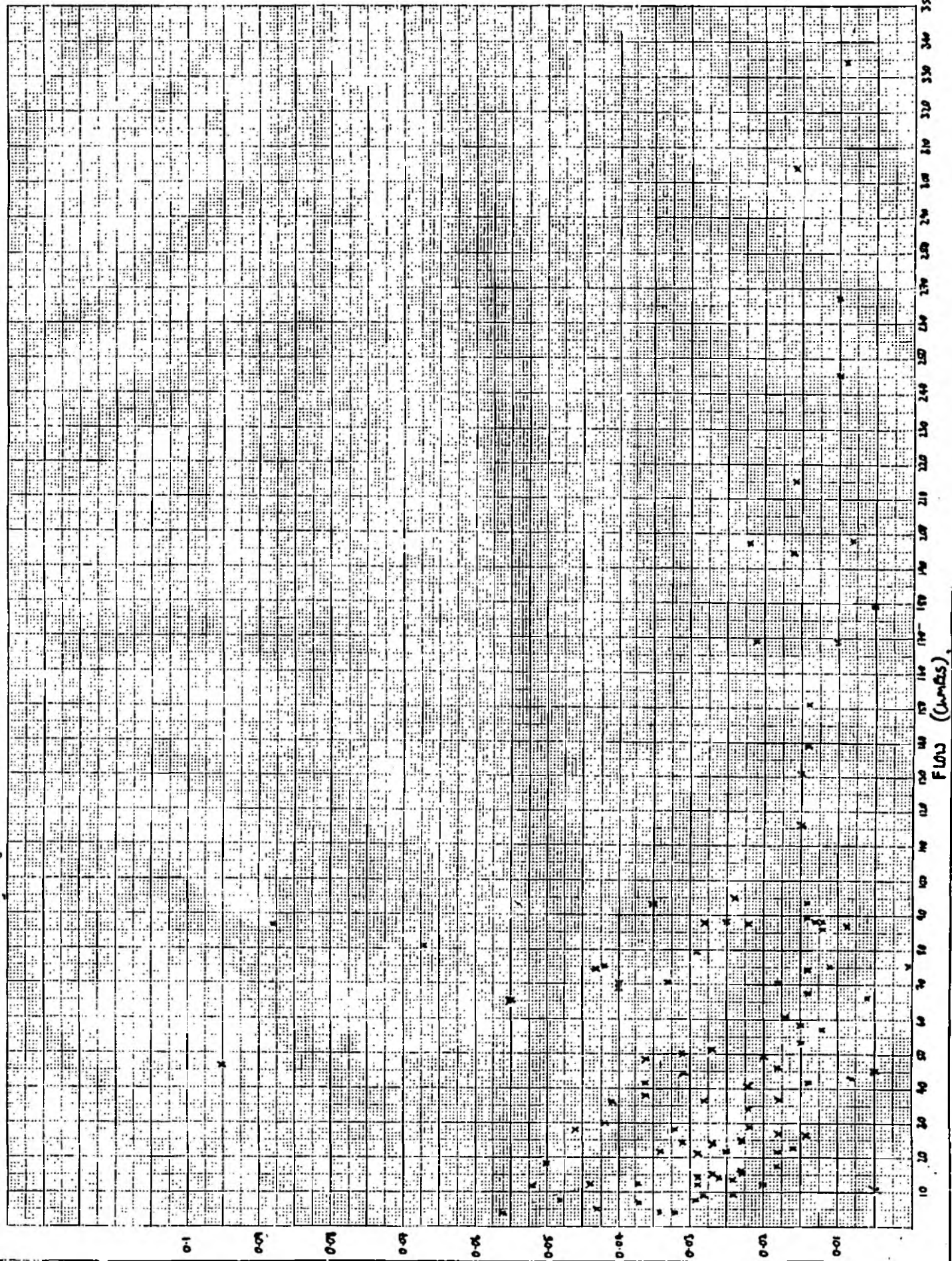
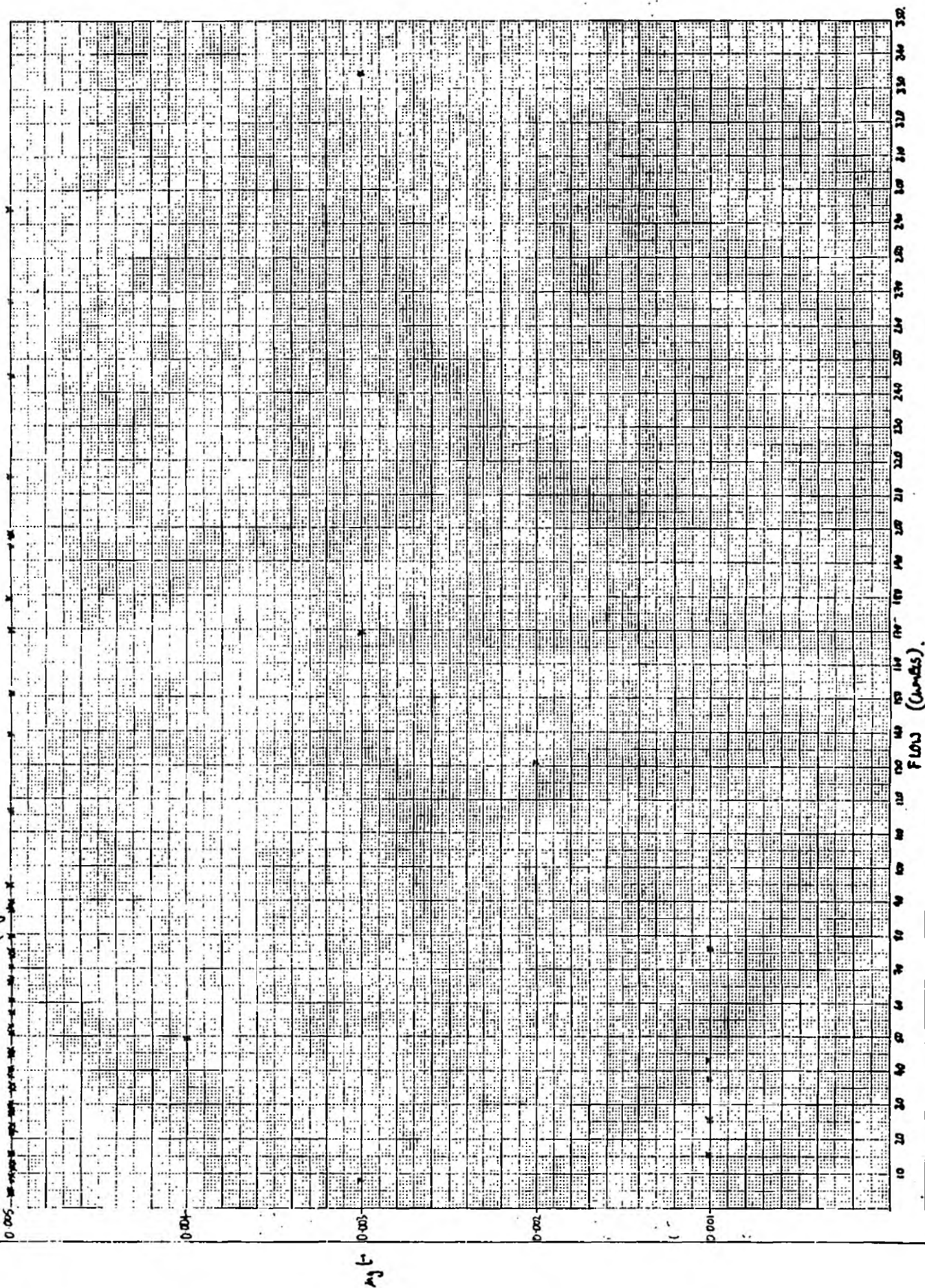


Fig 10

VARIATION OF DISCHARGE (CUMULATIVE) WITH RATE OF FLOW IN THE THAMES AT TROMPTON LANE (9/11/50 - 9/18/50)

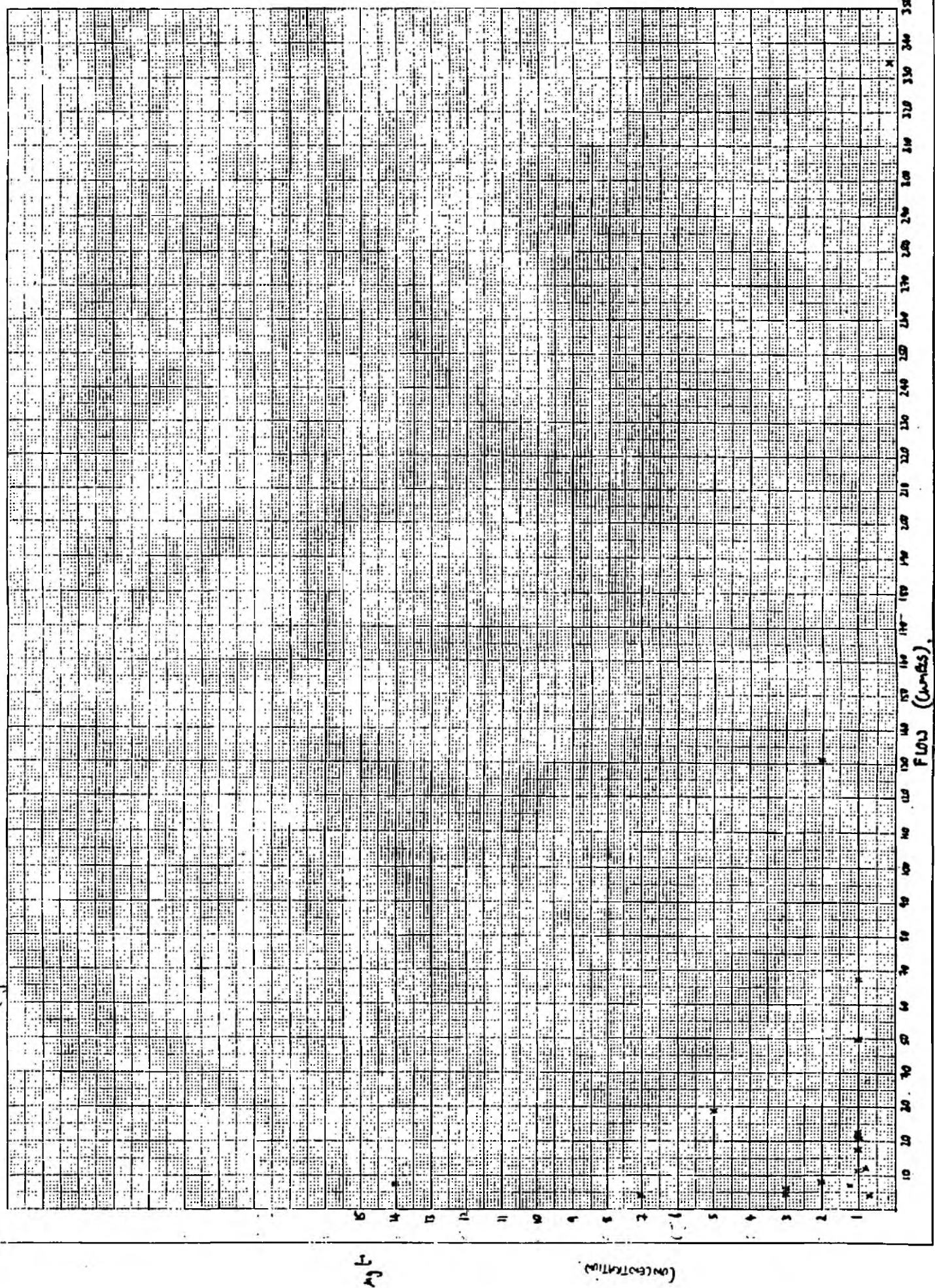
(Fig. 11)





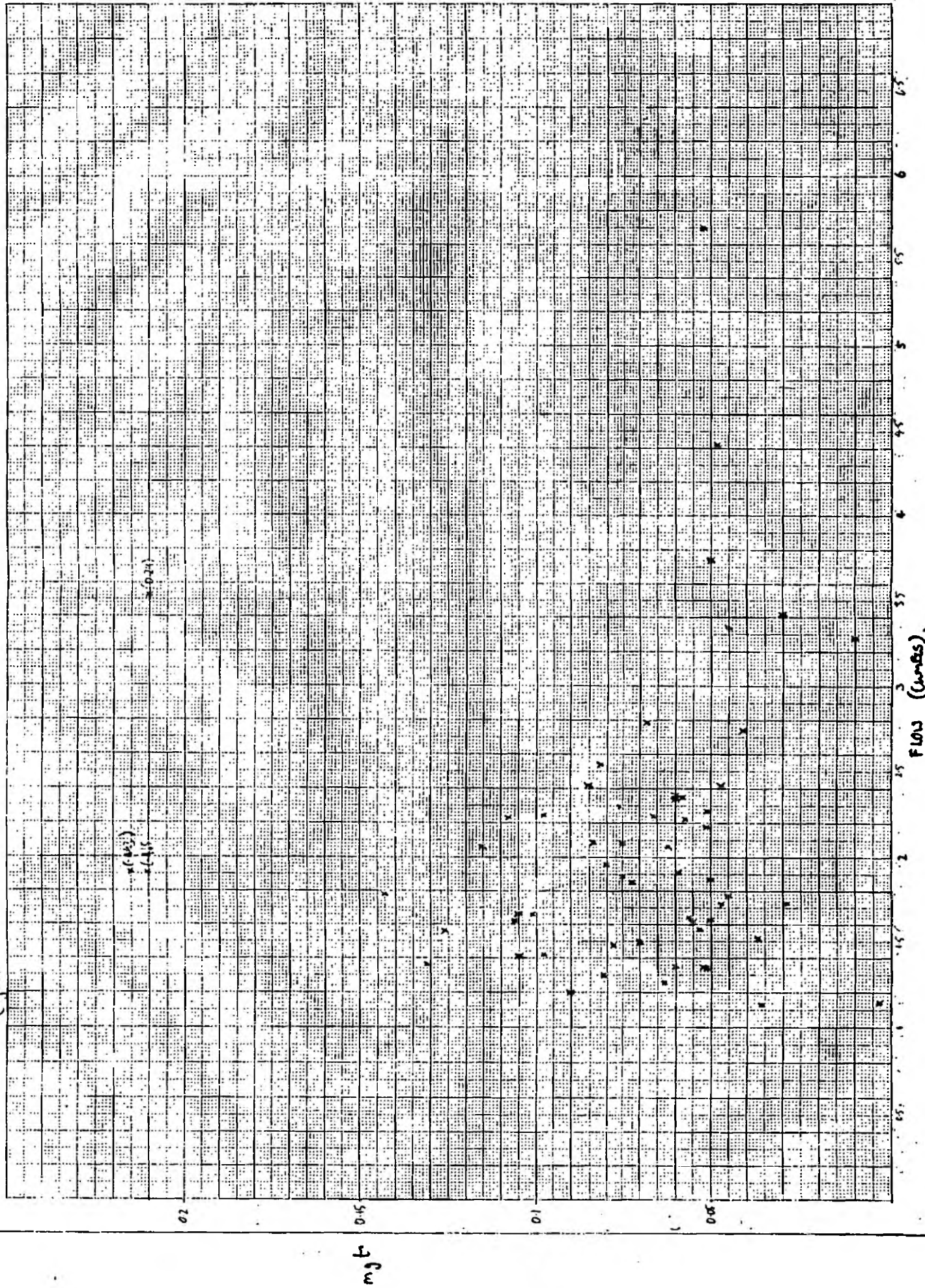
VARIATION OF TELLOURETHANE CONCENTRATION WITH FLOW IN THE TUNNELS AT TRODINGTON LK. (9/1/86 - 9/8/90)

(Fig. 12)



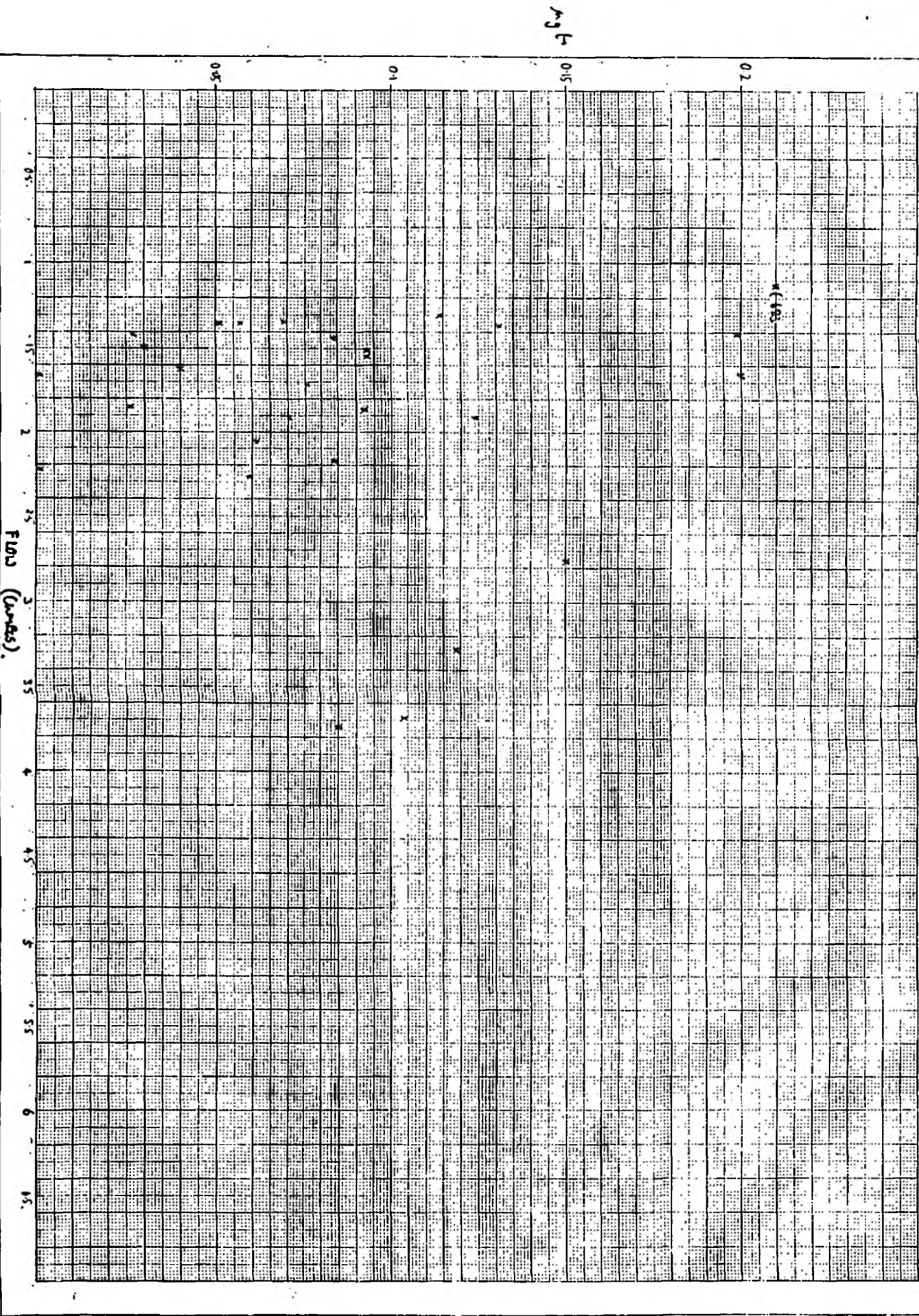
VARIATION OF TANK OPERATING WITH FLOW IN THE LANGLEY AT THE CAUSEWAY, DANDENBURGH (6288 - 12990).

(Fig. B)





Variation of H<sub>2</sub>O Specimen Concentration With Flow in the Waste at the Cassidy, Wabamun Unit (14-12-87 - 88/90)



Variation of Drag Coefficient ( $C_D$ ) with Flow in the Laminar Region (Fig. 15) (21-388-8842)

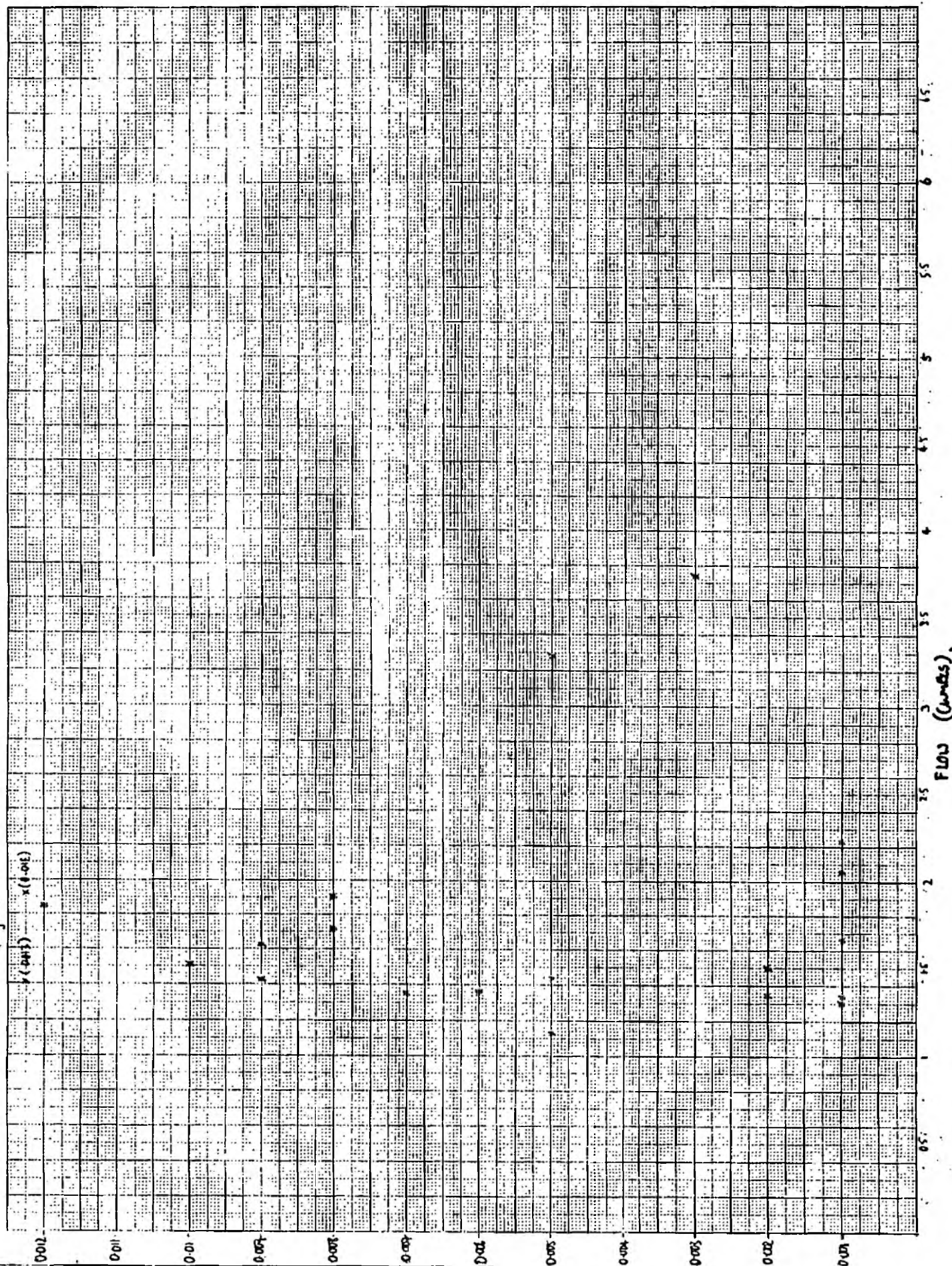
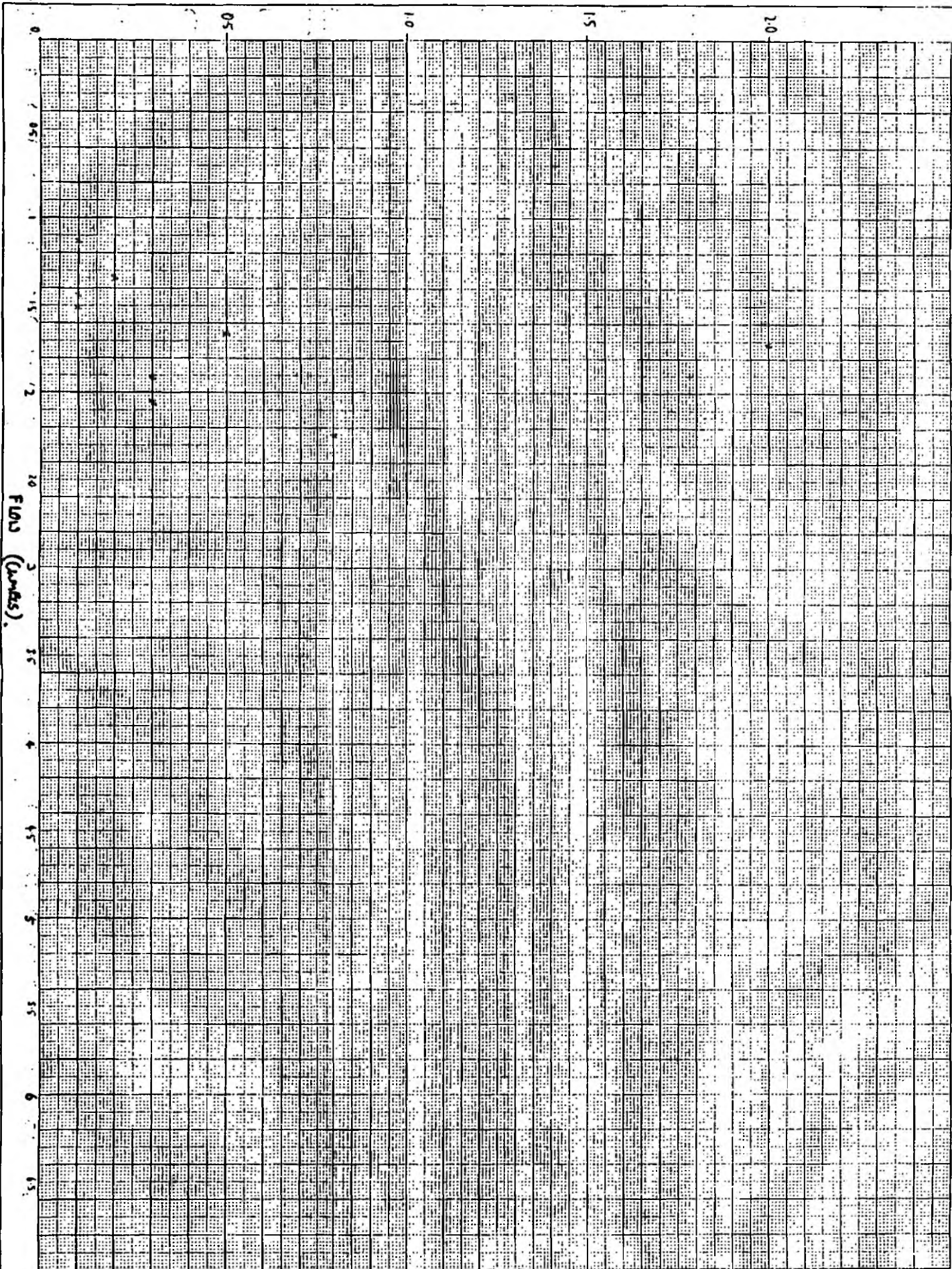


Fig 15

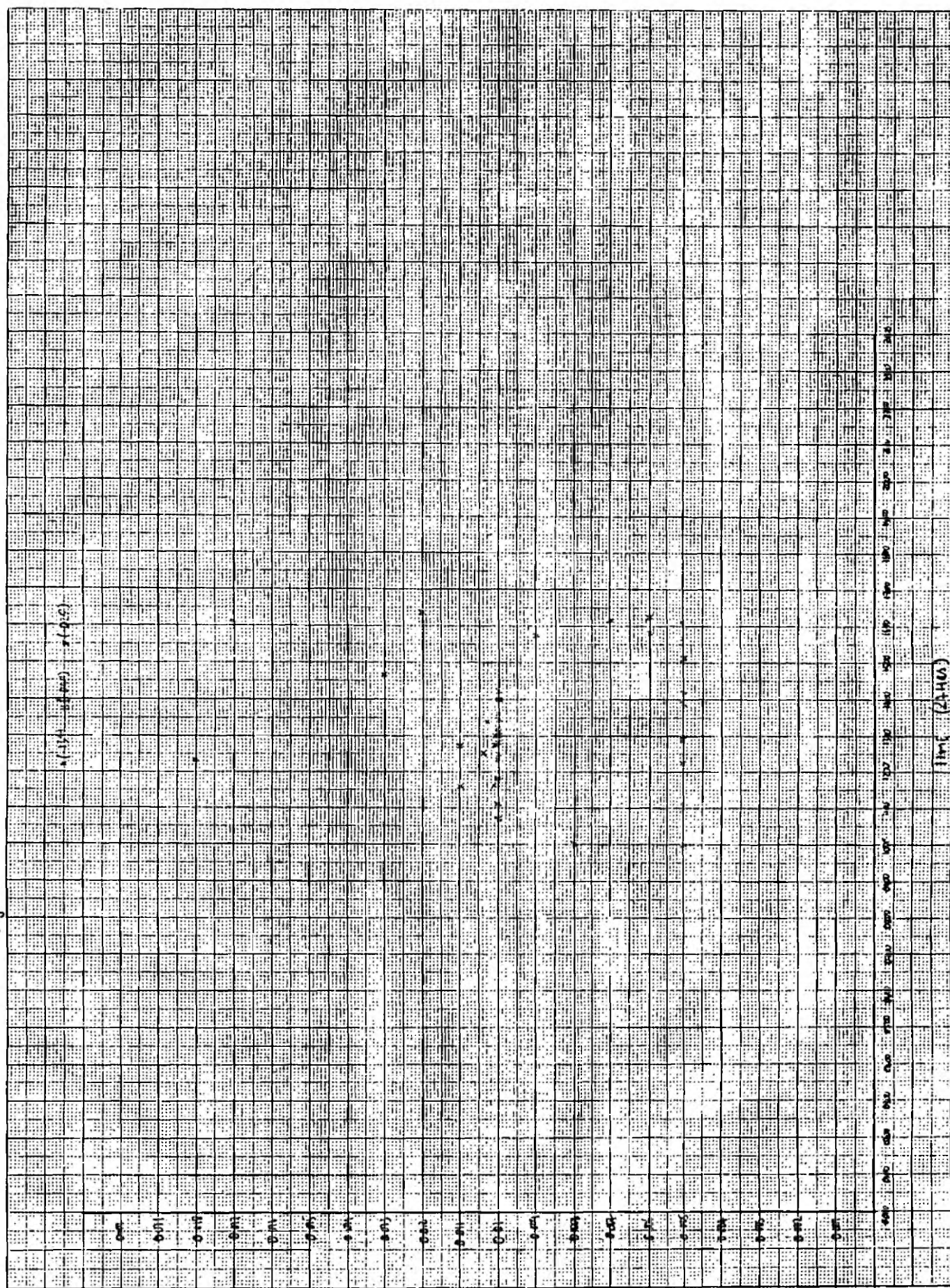


VARIATION OF TELLUROMETER CONCENTRATION WITH FLOW IN THE WHEEL AT THE (MSEURY, LINDSAY) (11-12-89 - 12-9-90)

(Fig 16)



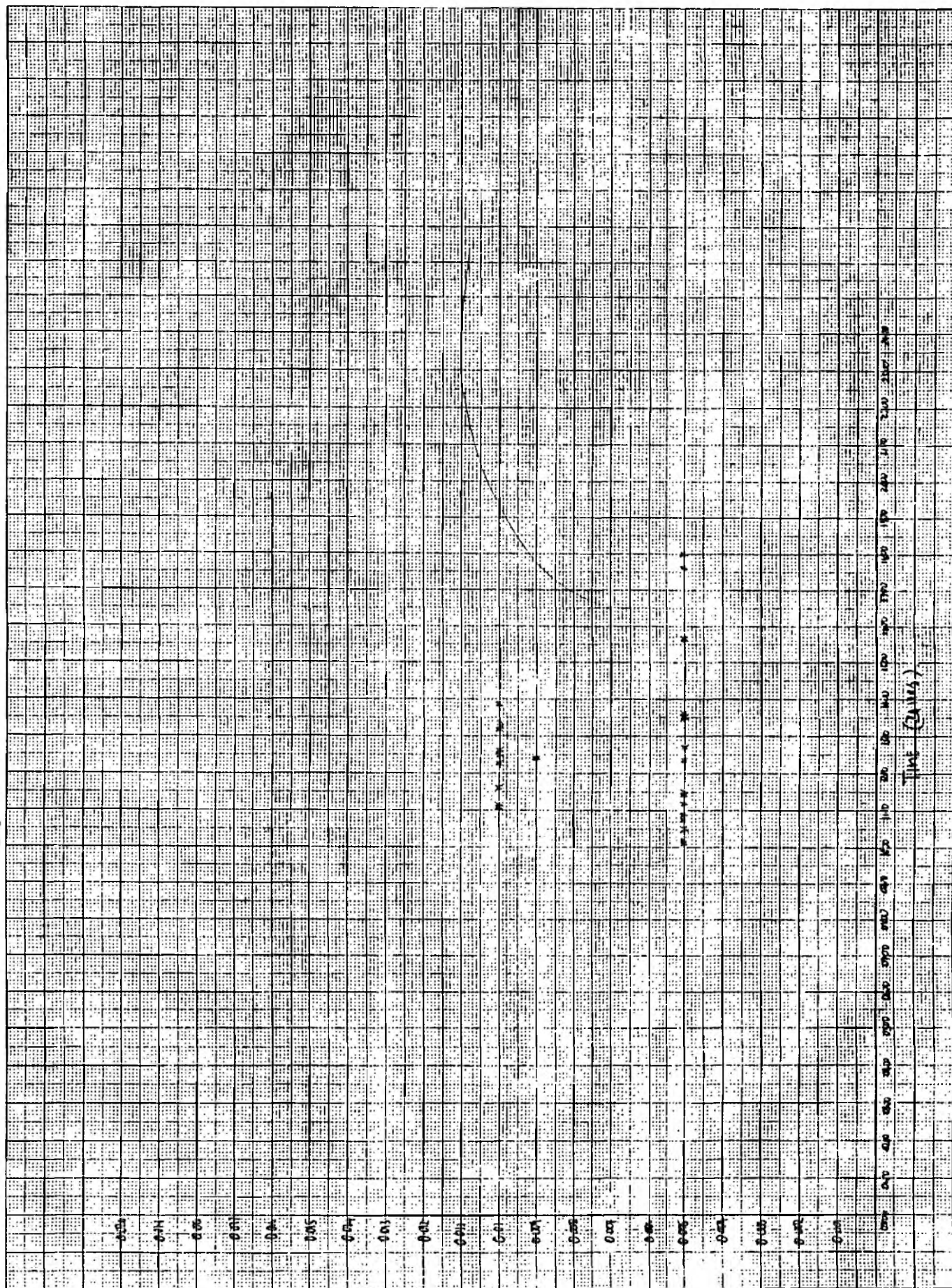
VARIATION IN NUCLEAR CONCENTRATION WITH STIMULUS TIME IN THE UMBILIC (S3-86-12-7-90)  
(Fig. 17)



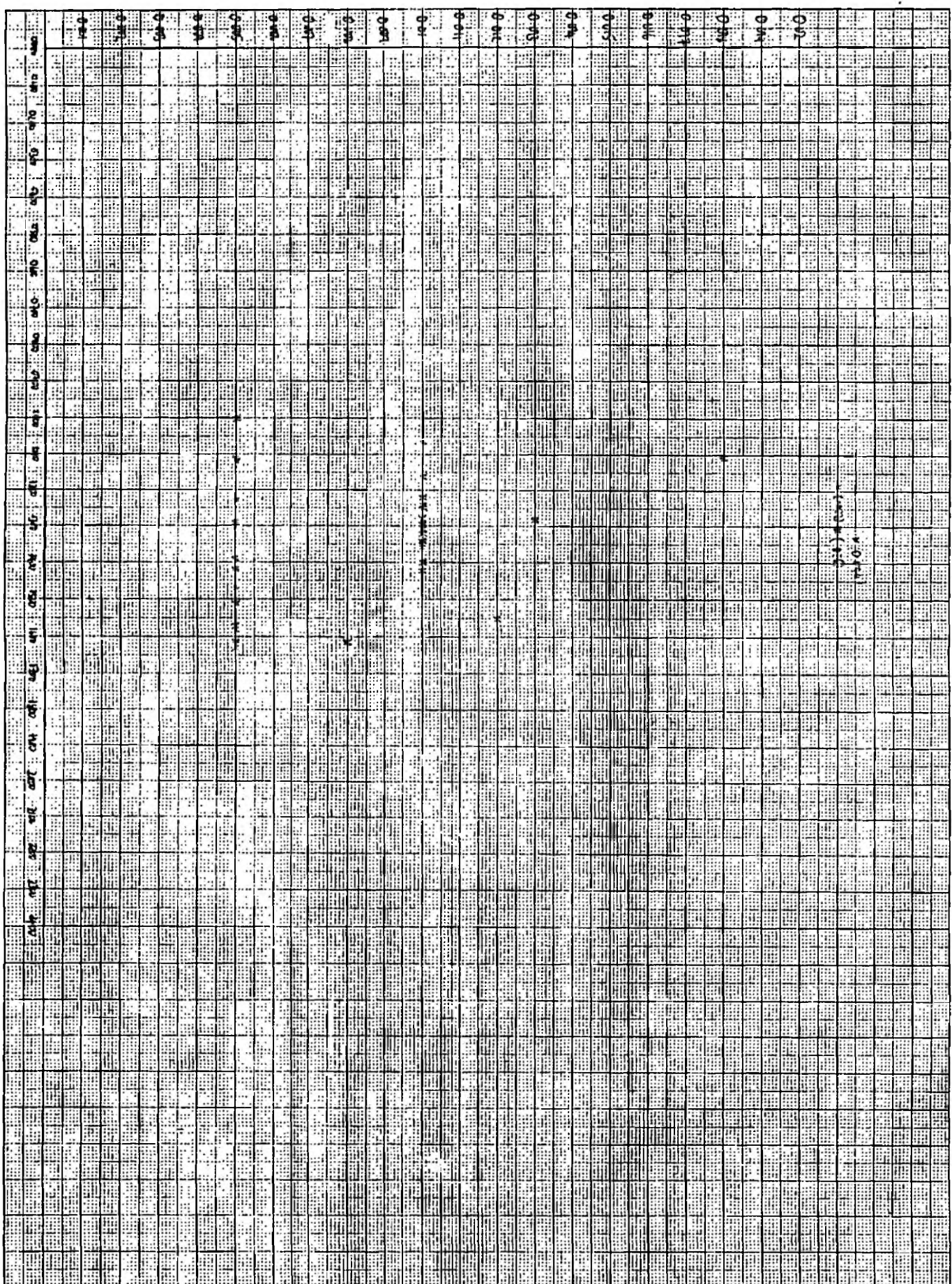


Variation in Maximum Contamination with Sampling Time in the Beaver Lake at Peoria, Illinois  
(11-3-86 - 12-9-90)

(Fig. 18)



Velocity in (Minimum Concentration) Data Station Time in the Limb at the (MAGNET) (53.80 - 12.40)

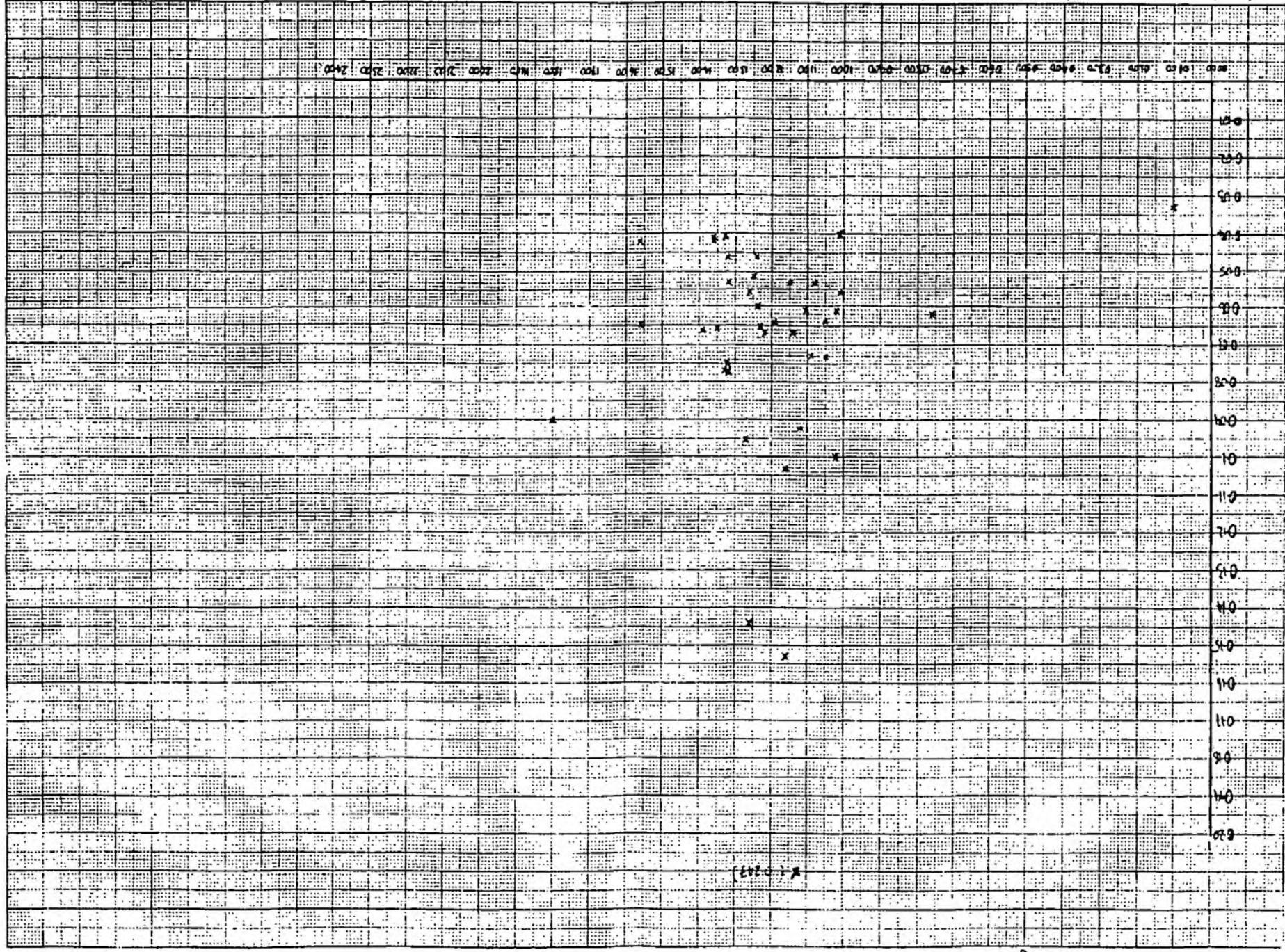


Time



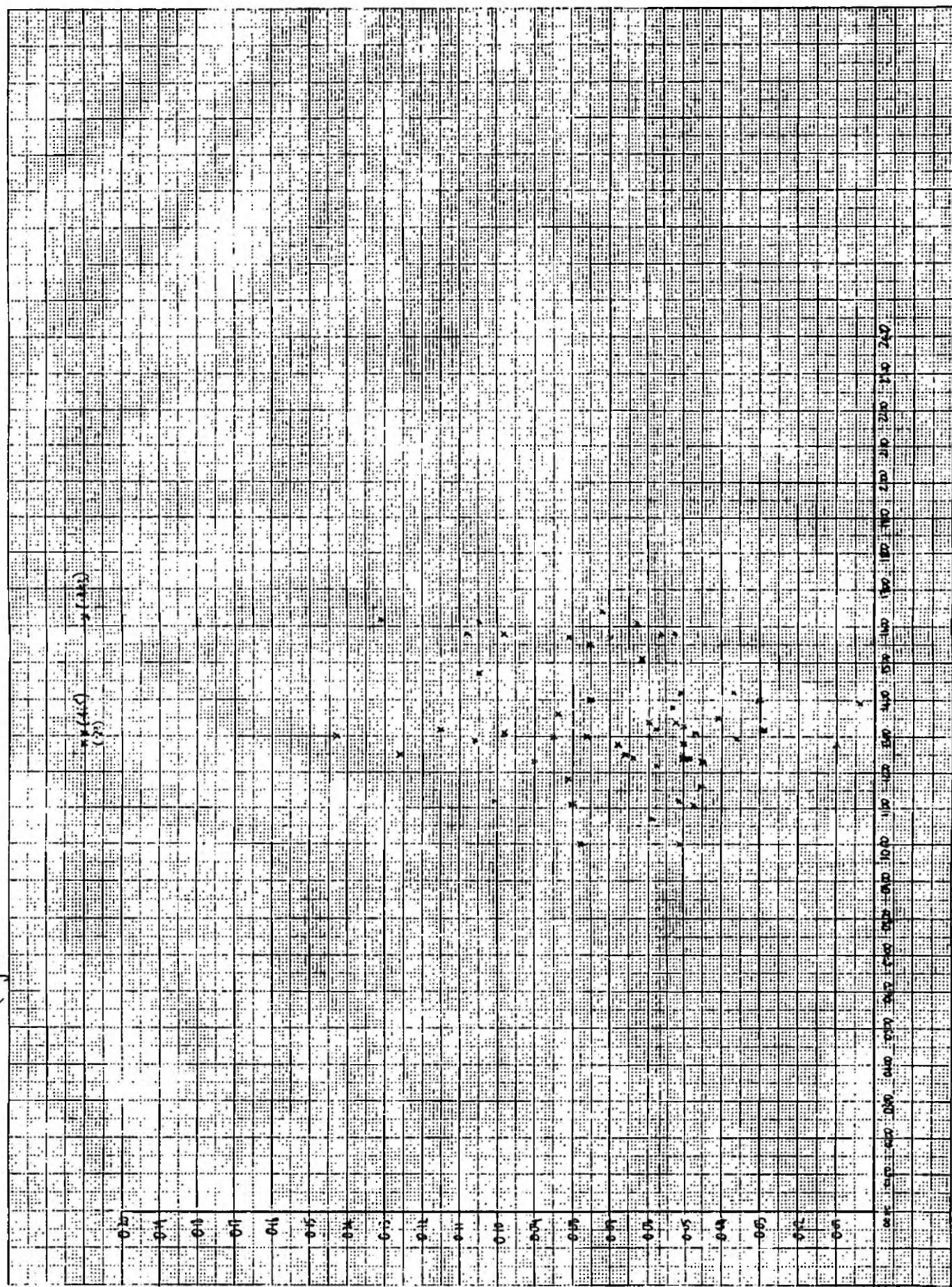
Variation in Line Concentration with Sampling Time in the Beverly Brook at FCI's 6604 (11-2-66 - 11-9-66)

(Fig 20)



mg/L

Variation in Zinc Concentration With Sampling Time in the Wangle at the Causeway (6288 - 12990)  
 (Fig 21)



Time (hours)



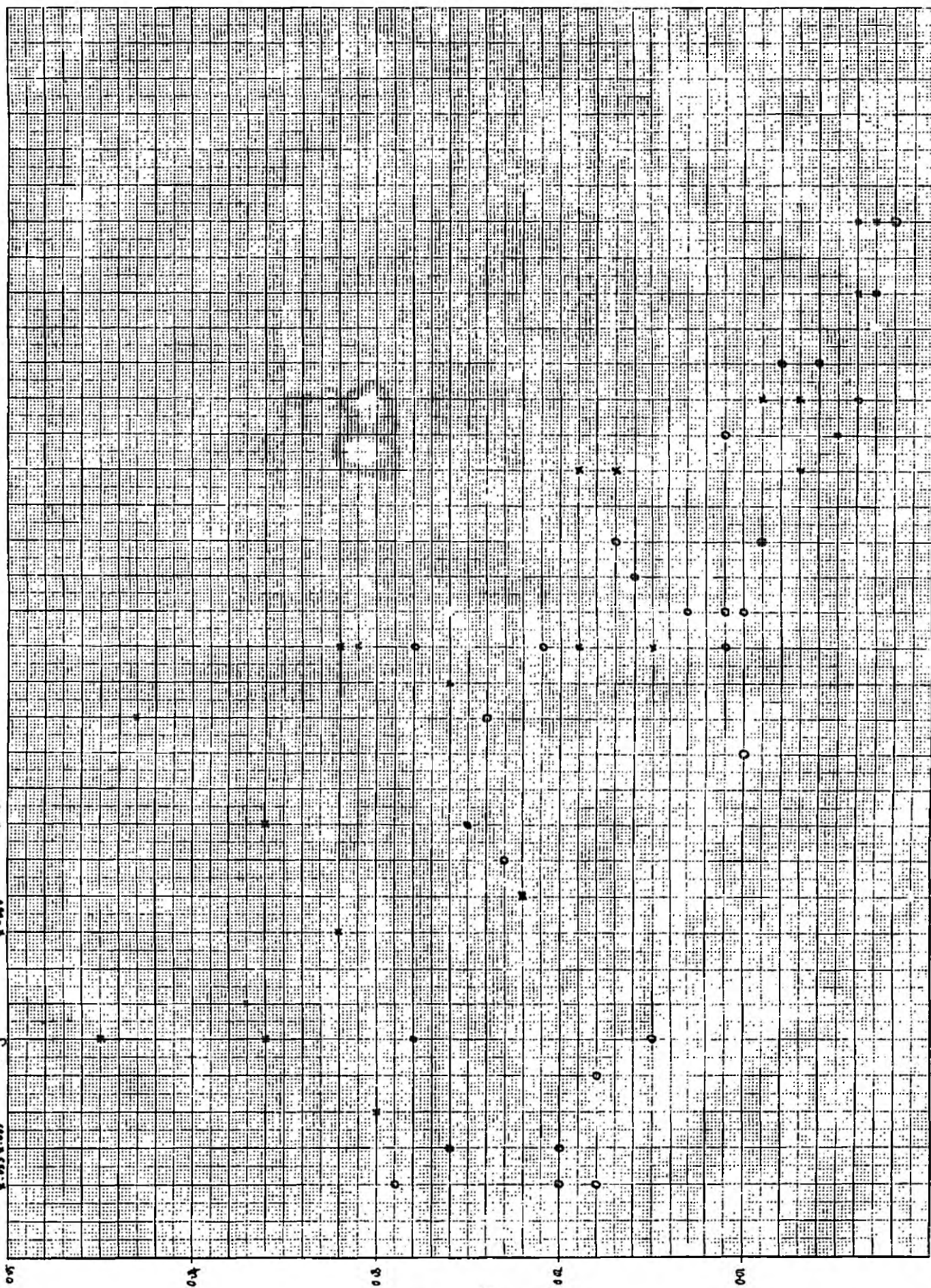
VARIOUS IN HIGH Y CONCENTRATION THROUGHOUT THE THINER ZONES AND IN RELATION TO FLOW RATES AT TEMPERATURE (1 unit below sampling).

(Fig. 22)

$\times 10^3$   
 $\times 10^3 \pm 0.05$

$\times 10^3$   
 $\times 10^3 \pm 0.05$

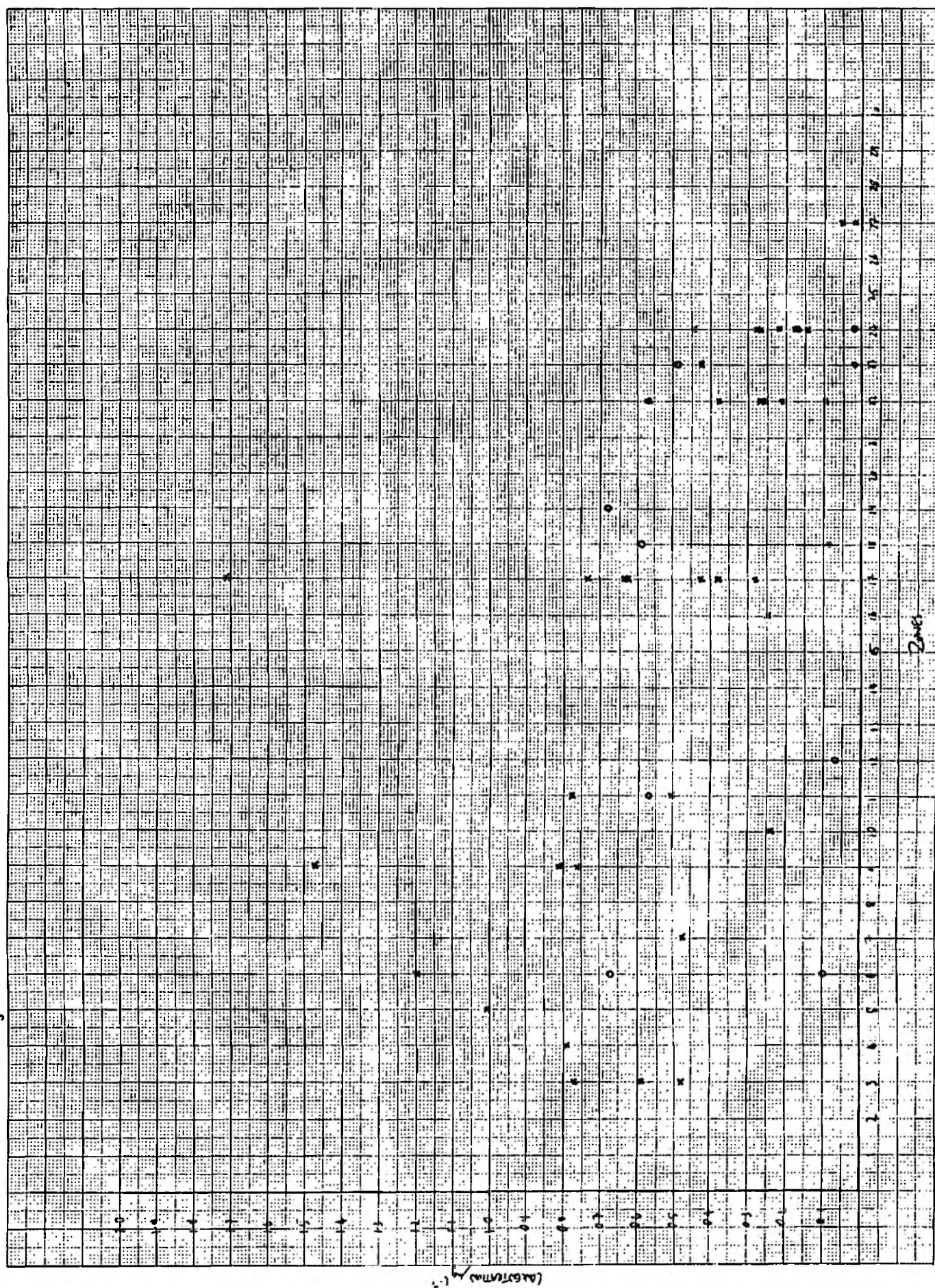
$\times 10^3$



ZONES

0.000000 - OVER 100  
0.000000 - UNDER 100

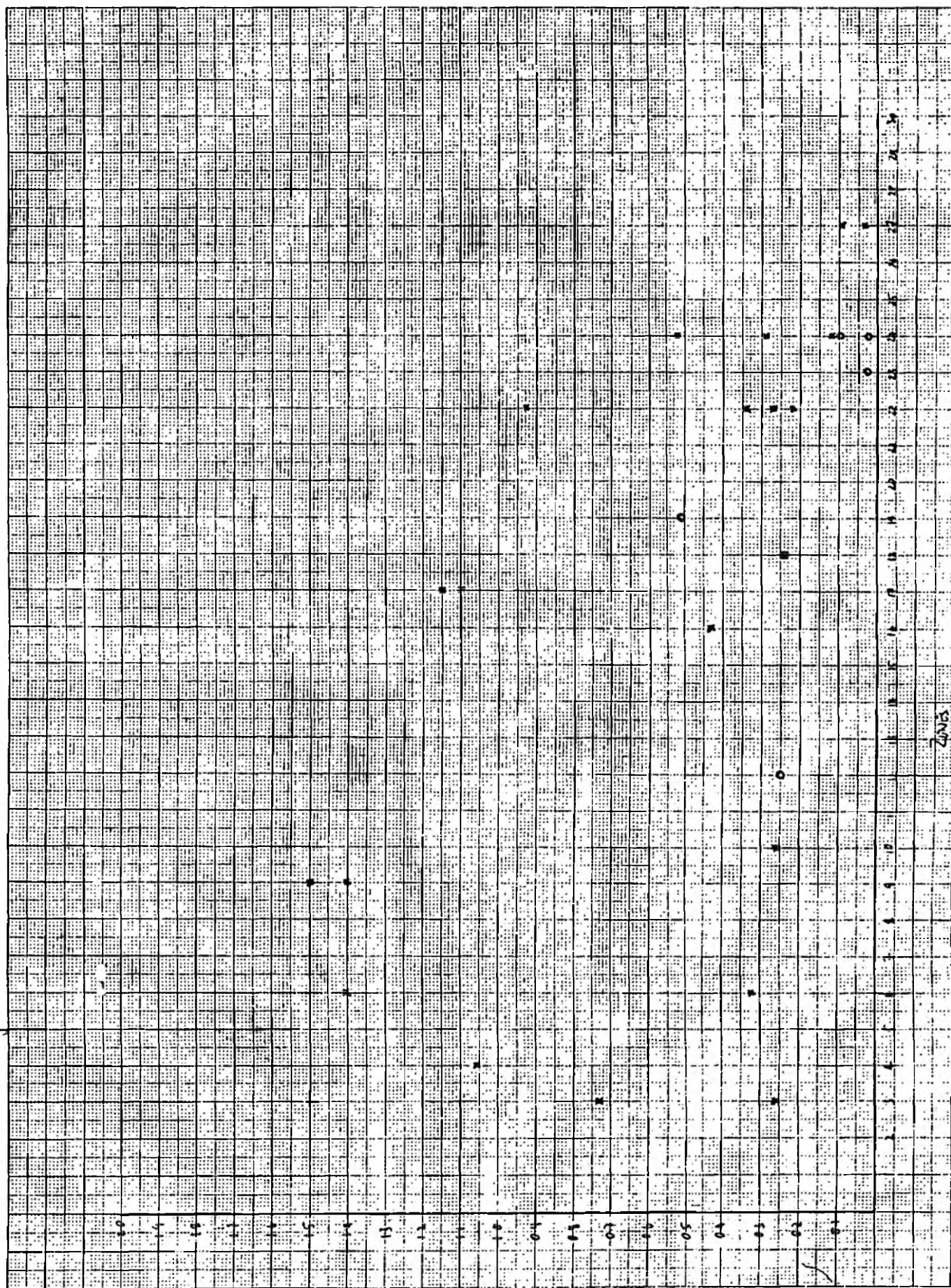
Variation in Aggregate Concentration Throughout the Times Zones in Relation to Peak Rates at Termination Life  
 Fig. 23.



O 240 (times)  
 X 240 (times)



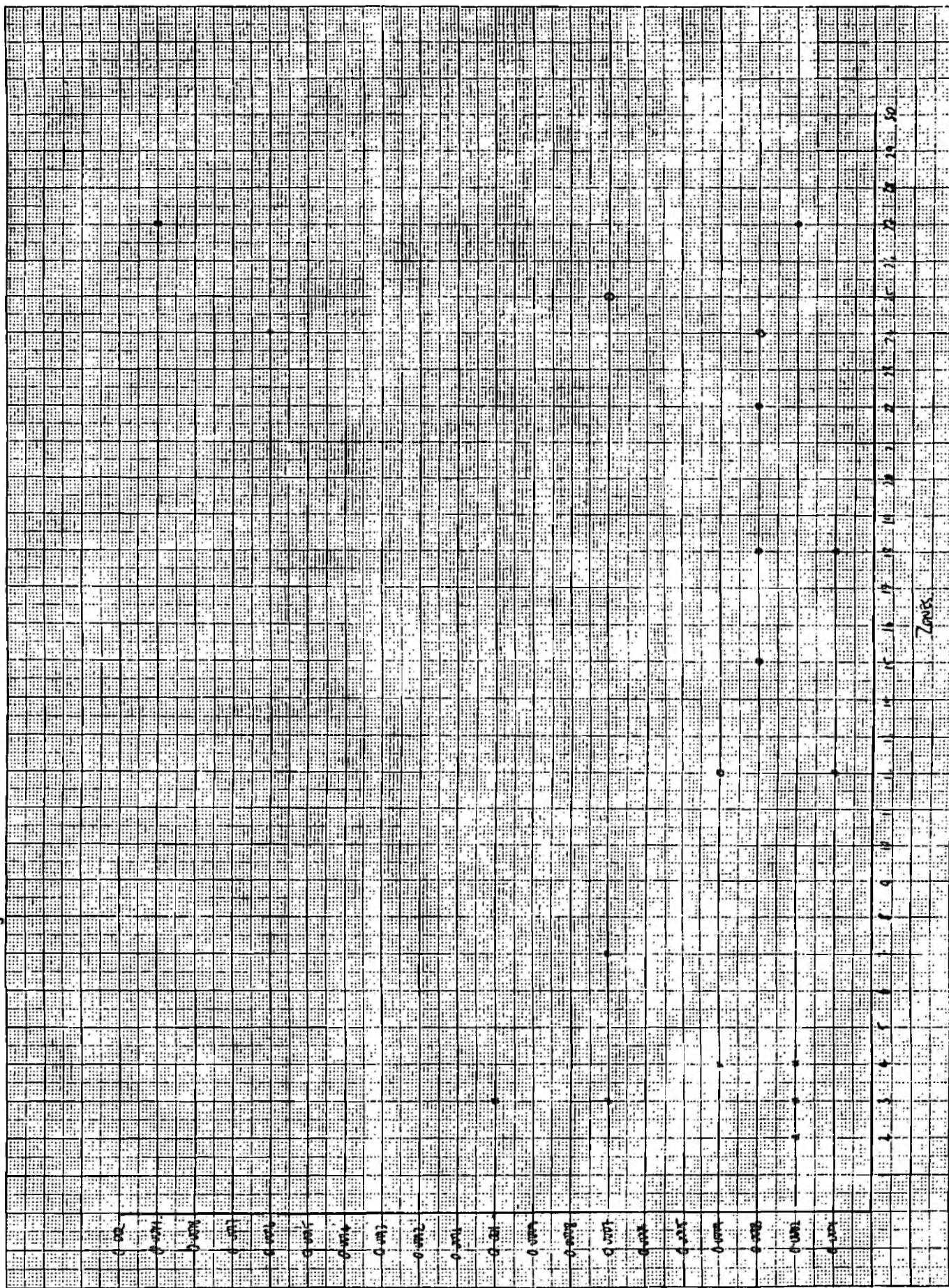
Fig 24 VARIATION IN SINGLED CONCENTRATION THROUGHOUT THE TURNING ZONE AND IN RELATION TO FLOW RATES AT TERMINATION



0.250 (units)  
x 4.20 (units)

2003

Variation in  $\epsilon_{\text{H}}^{\text{H}}$  (Average) (Calculation in the present case is based on the relationship at equilibrium)



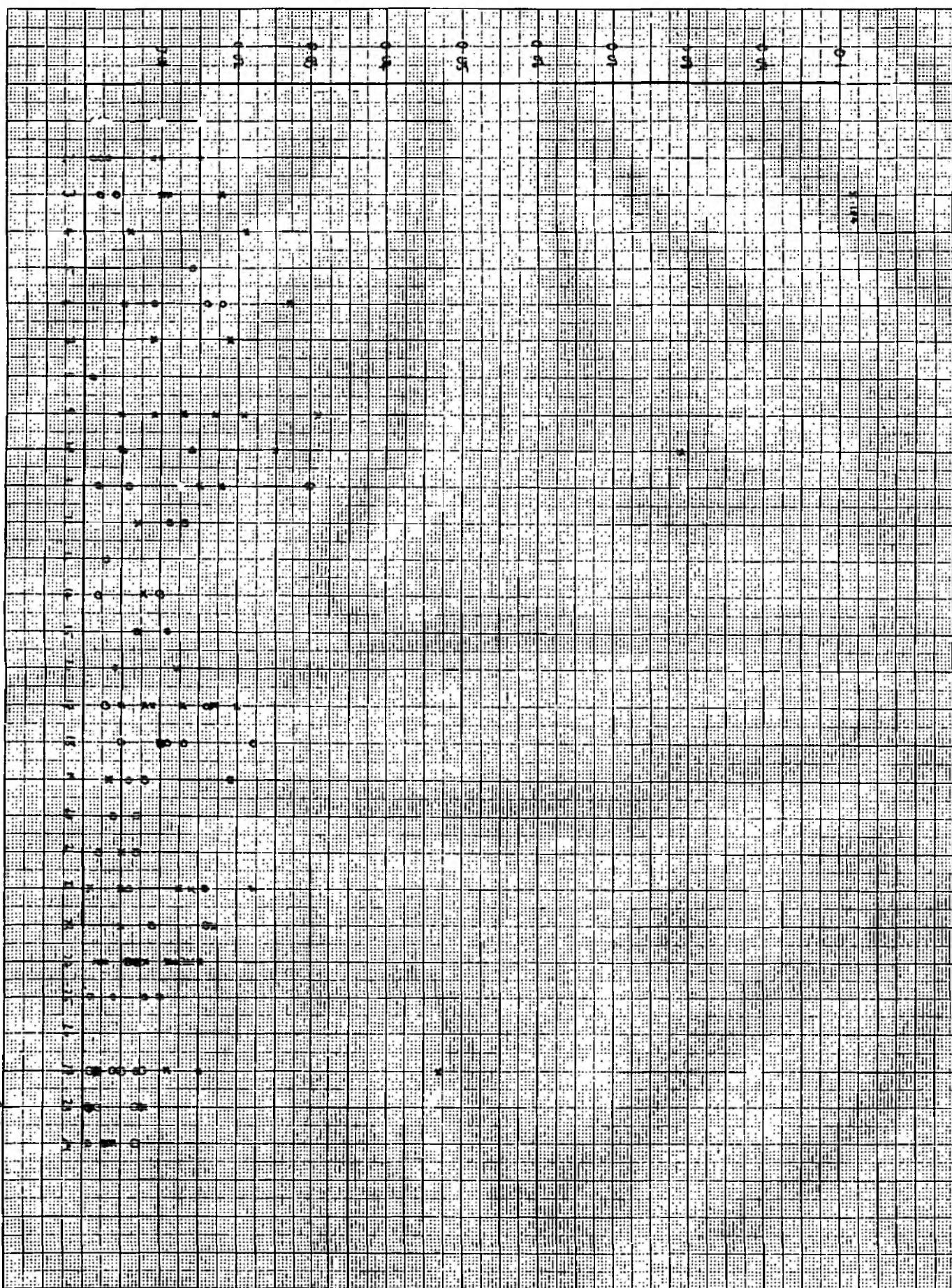
0 > 40 (units)  
x < 20 (units)



Location in lower concentration in the thinnest zones in relation to flow units at Tetonville, Va.

Fig. 26

1947

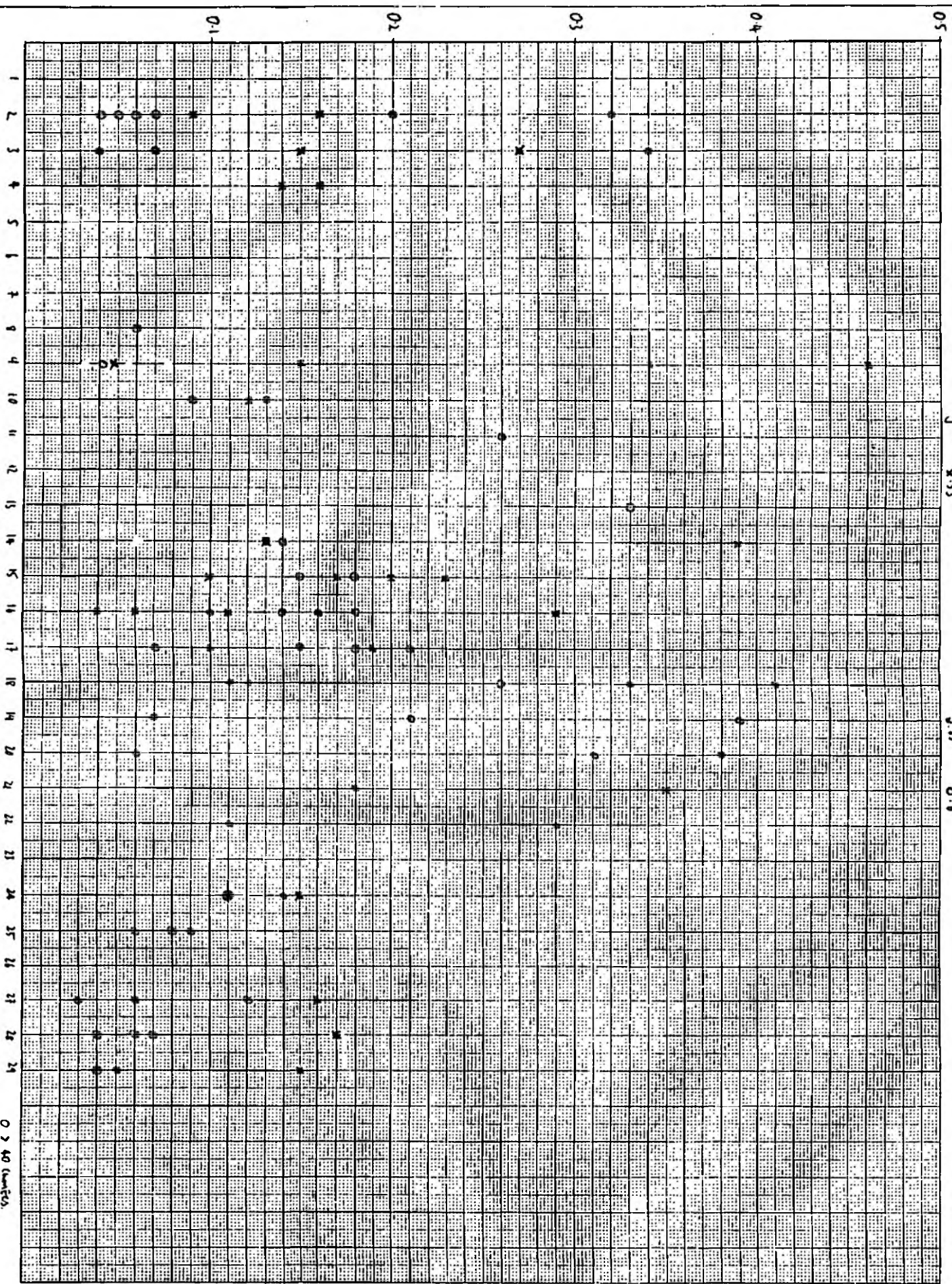


Zones

0.050 units  
x 420 units

Number in Number Conversions in The Terms Table in Relation to Fractions at Transition

Fig 1



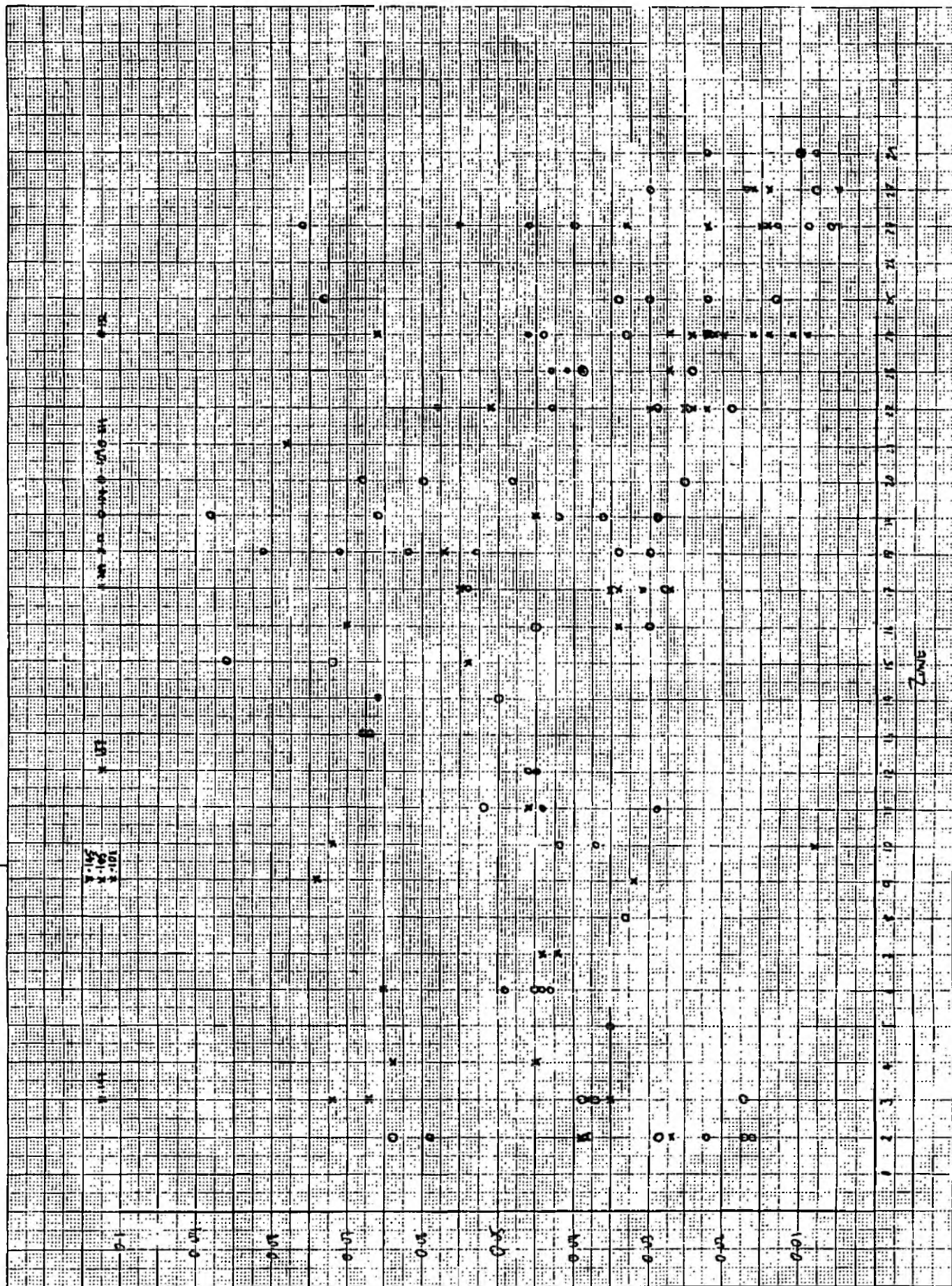
Lambdas

0 > 40 (units)  
x < 29 (units)



Variation in Zinc Concentration Throughout the Turners Zoned in Relation to Fly Ashes at Thompson.

Fig. 28.



○ Zinc (ppm)  
× Fly Ashes (meters)